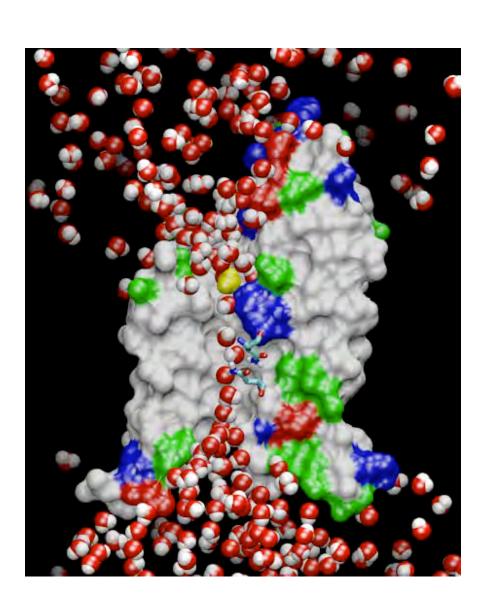
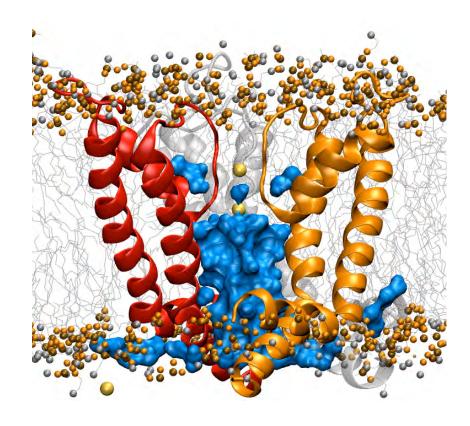
Investigating Biological Membranes and Membrane Proteins Using Advanced Simulation Technologies





Emad Tajkhorshid

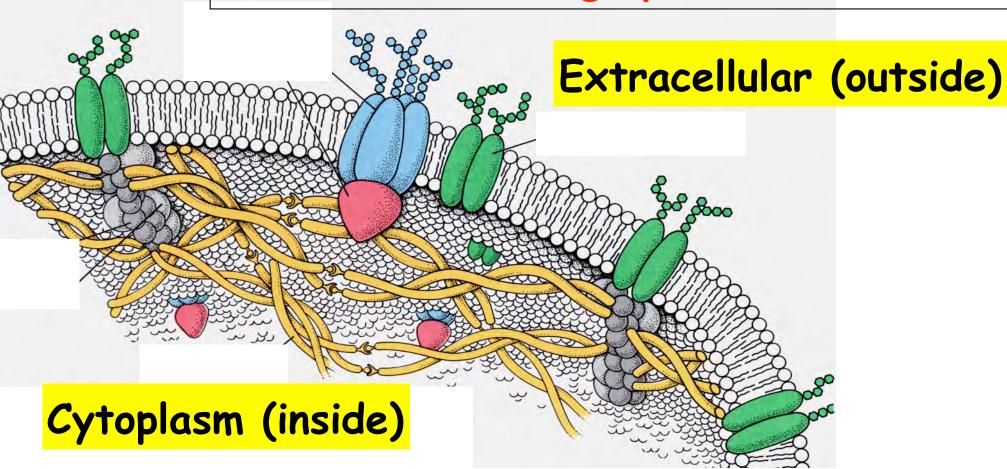
NIH Center for Macromolecular Modeling and Bioinformatics, Beckman Institute, University of Illinois at Urbana-Champaign



Why Do Living Cells Need Membrane

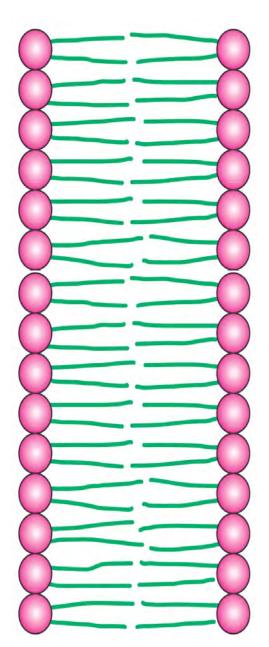
 Living cells also need to exchange materials and information with the outside world

... however, in a highly selective manner.

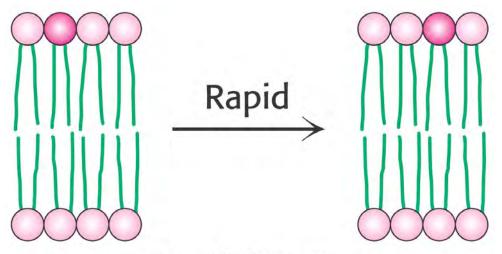


Phospholipid Bilayers Are Excellent Materials For Cell Membranes

- · Hydrophobic interaction is the driving force
- Self-assembly in water
- Tendency to close on themselves
- Self-sealing (a hole is unfavorable)
- Extensive: up to millimeters



Lipid Diffusion in a Membrane

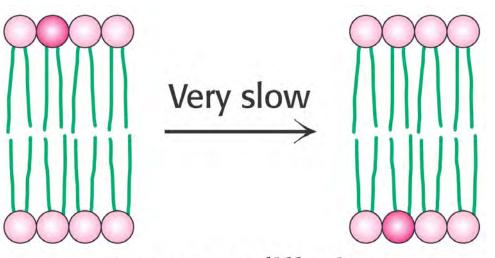


Lateral diffusion

$$D_{lip} = 10^{-8} \text{ cm}^2.\text{s}^{-1}$$

$$(50 \text{ Å in } \sim 5 \times 10^{-6} \text{ s})$$

 $D_{wat} = 2.5 \times 10^{-5} \text{ cm}^2.\text{s}^{-1}$ Modeling mixed lipid bilayers!



Tranverse diffusion (flip-flop)

Once in several hours!

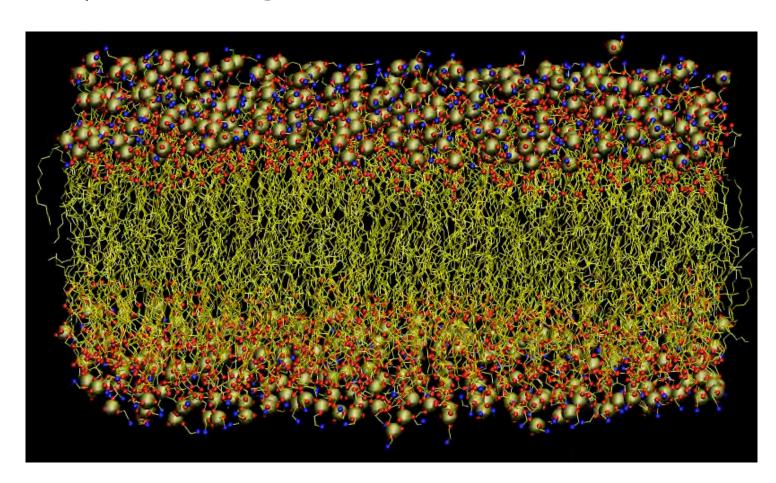
 $(\sim 50 \text{ Å in} \sim 10^4 \text{ s})$

~9 orders of magnitude slower ensuring bilayer asymmetry

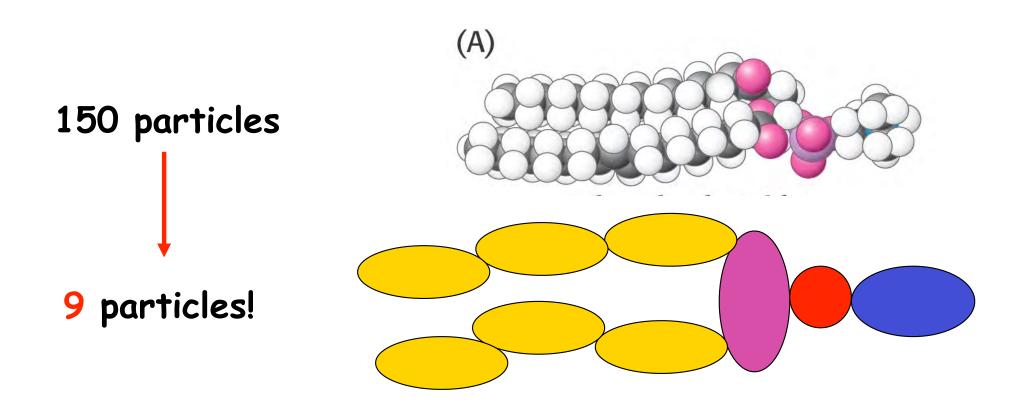
Technical difficulties in Simulations of Biological Membranes

- Time scale
- Heterogeneity of biological membranes

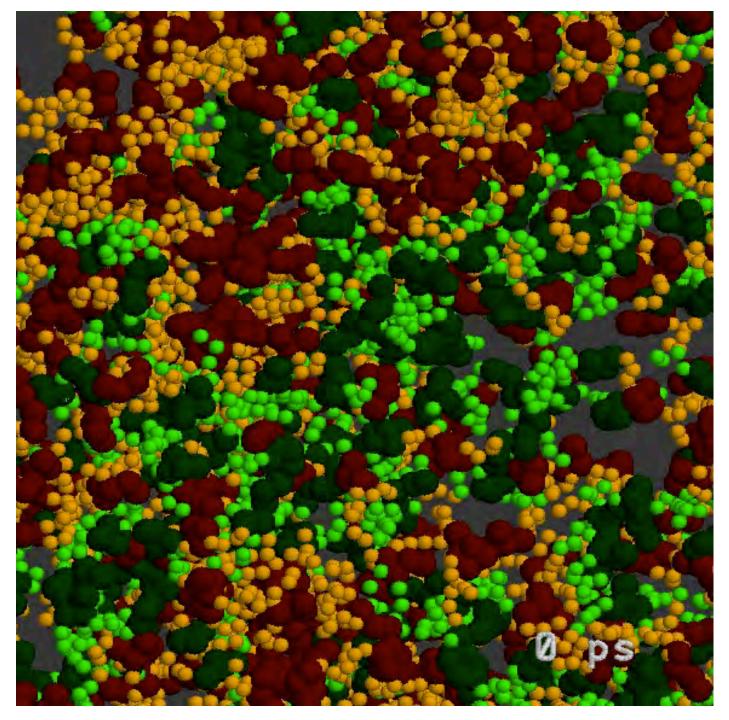
60 x 60 Å
Pure POPE
5 ns
~100,000
atoms



Coarse-grained modeling of lipids



Also, increasing the time step by orders of magnitude.



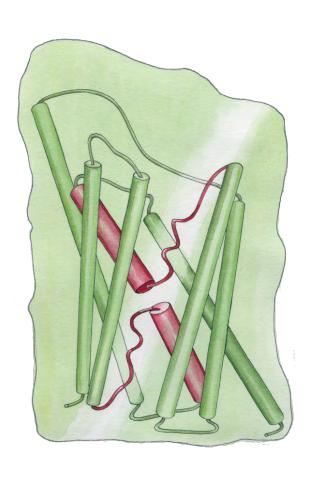
by: J. Siewert-Jan Marrink and Alan E. Mark, University of Groningen, The Netherlands

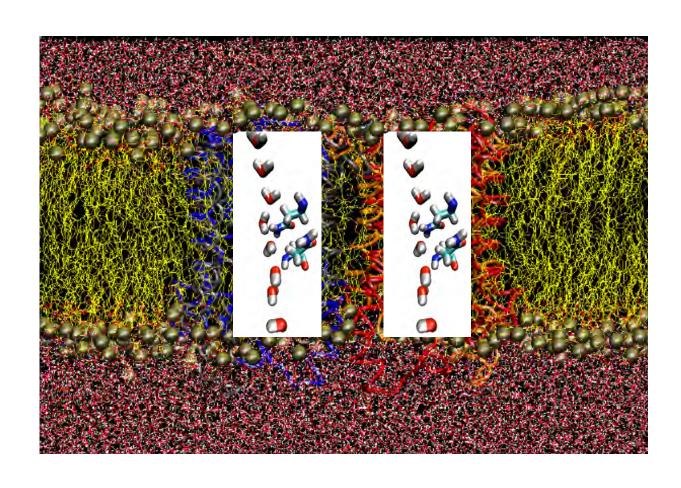
Analysis of Molecular Dynamics Simulations of Biomolecules

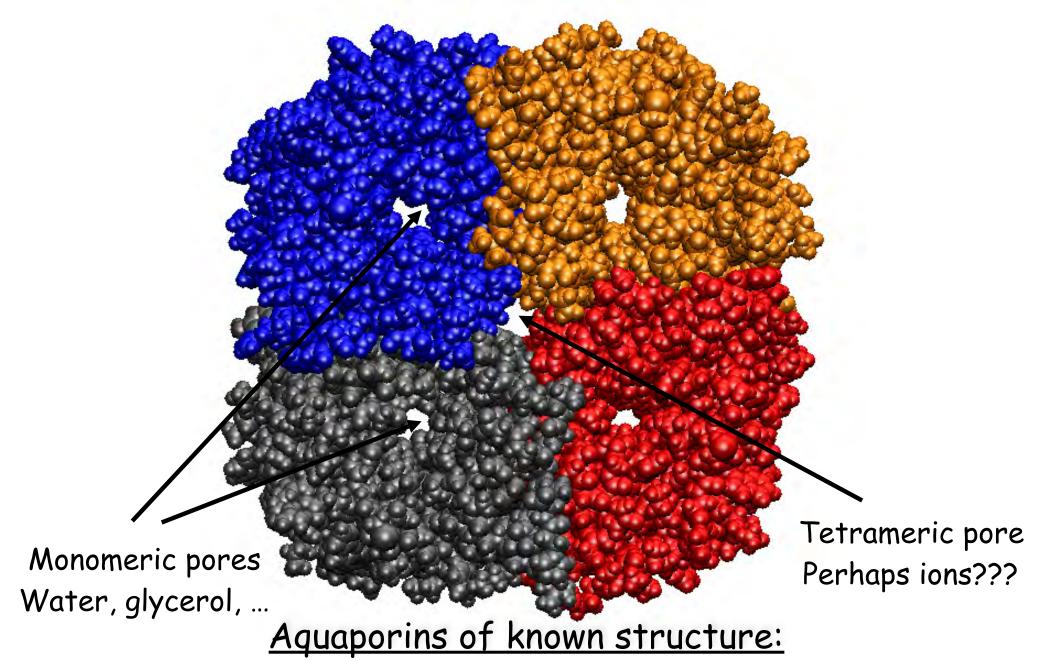
- A very complicated arrangement of hundreds of groups interacting with each other
- Where to start to look at?
- What to analyze?
- · How much can we learn from simulations?

It is very important to get acquainted with your system

Aquaporins





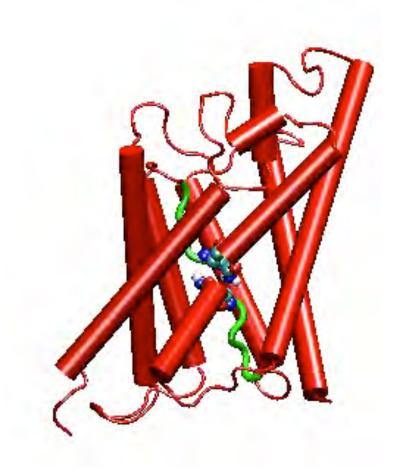


GlpF - E. coli glycerol channel (aquaglycerolporin)

AQP1 - Mammalian aquaporin-1 (pure water channel)
AqpZ and AQP0 (2004)

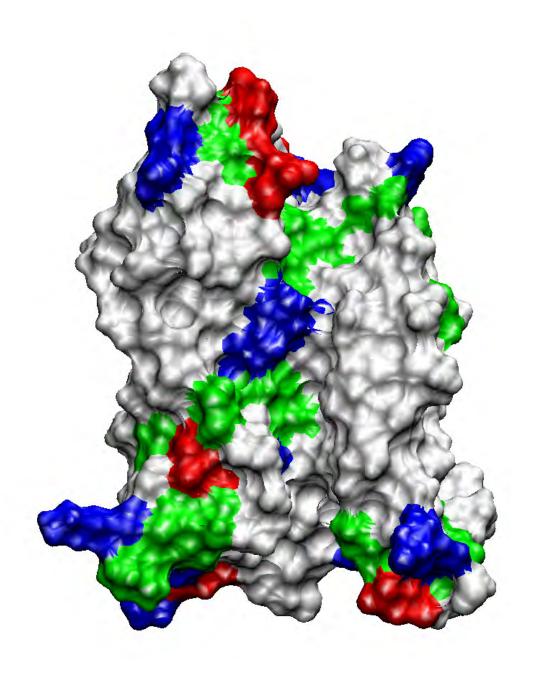
Functionally Important Features

- Tetrameric architecture
- Amphipatic channel interior
- Water and glycerol transport
- Protons, and other ions are excluded
- Conserved asparagine-prolinealanine residues; NPA motif
- Characteristic half-membrane spanning structure





A Semi-hydrophobic channel



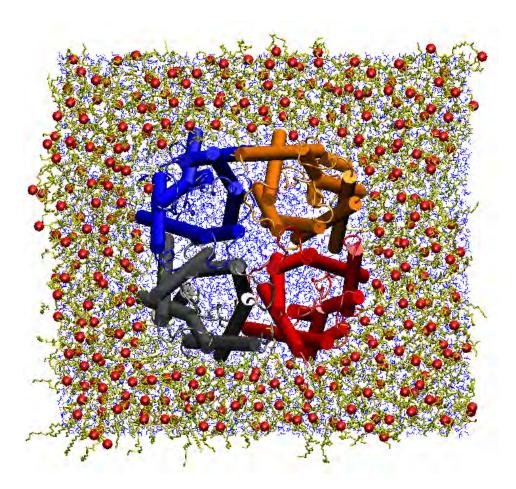
Molecular Dynamics Simulations

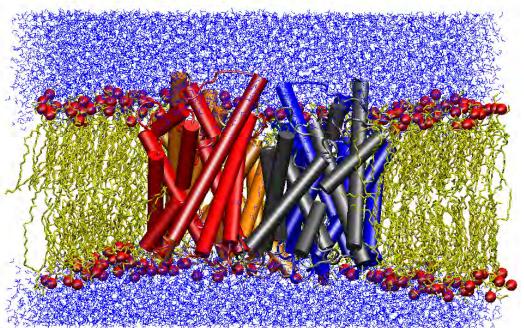
Protein: ~ 15,000 atoms

Lipids (POPE): ~ 40,000 atoms

Water: ~ 51,000 atoms

Total: ~ 106,000 atoms





NAMD, CHARMM27, PME

NpT ensemble at 310 K

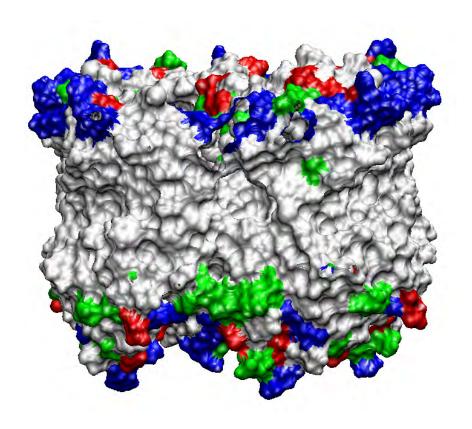
1ns equilibration, 4ns production

10 days /ns - 32-proc Linux cluster

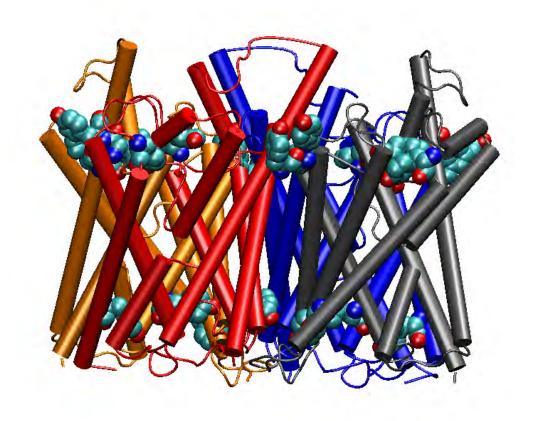
3.5 days/ns - 128 O2000 CPUs

0.35 days/ns - 512 LeMieux CPUs

Protein Embedding in Membrane

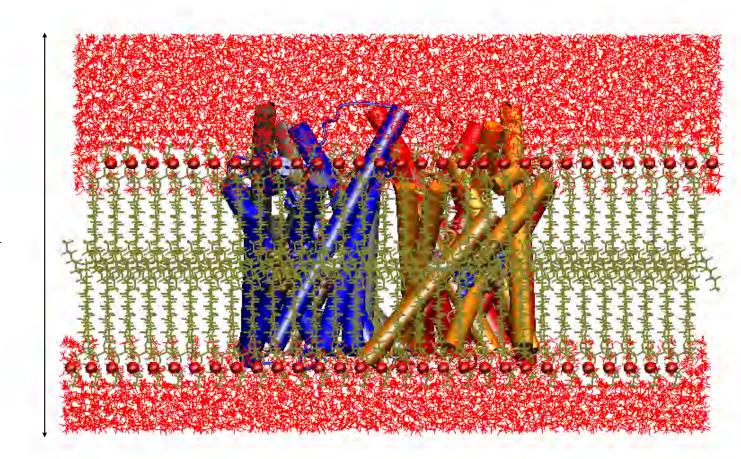


Hydrophobic surface of the protein



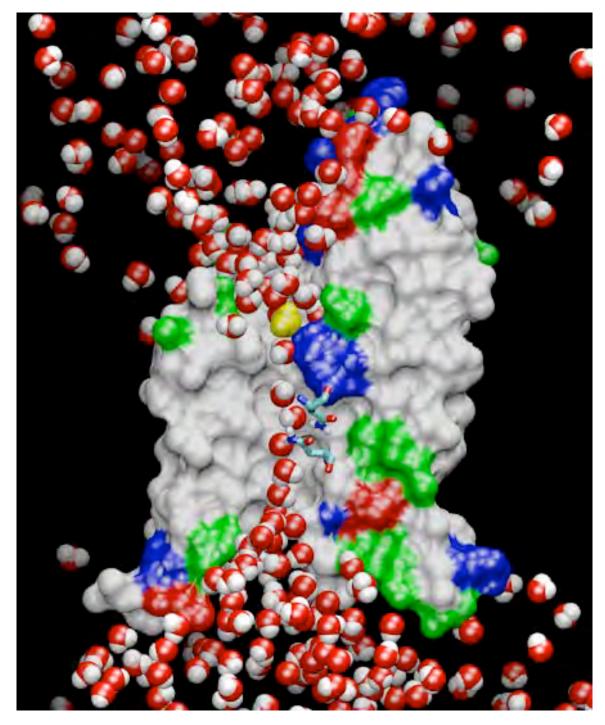
Ring of Tyr and Trp

Embedding GlpF in Membrane

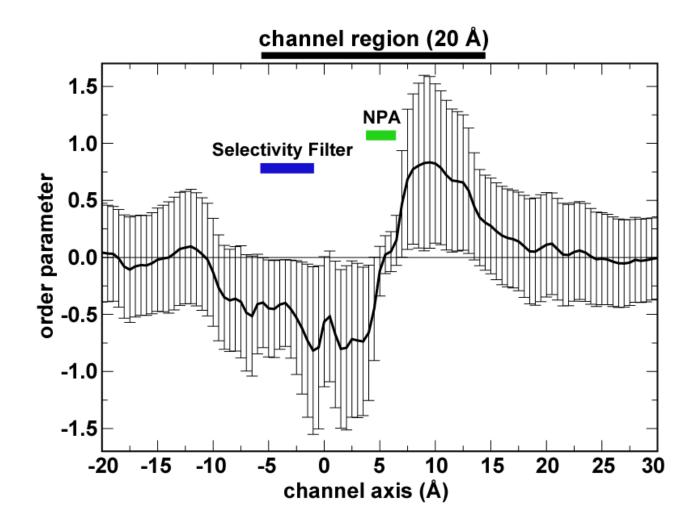


77 A

CHARMM-GUI



Animation available at the Nobel web site

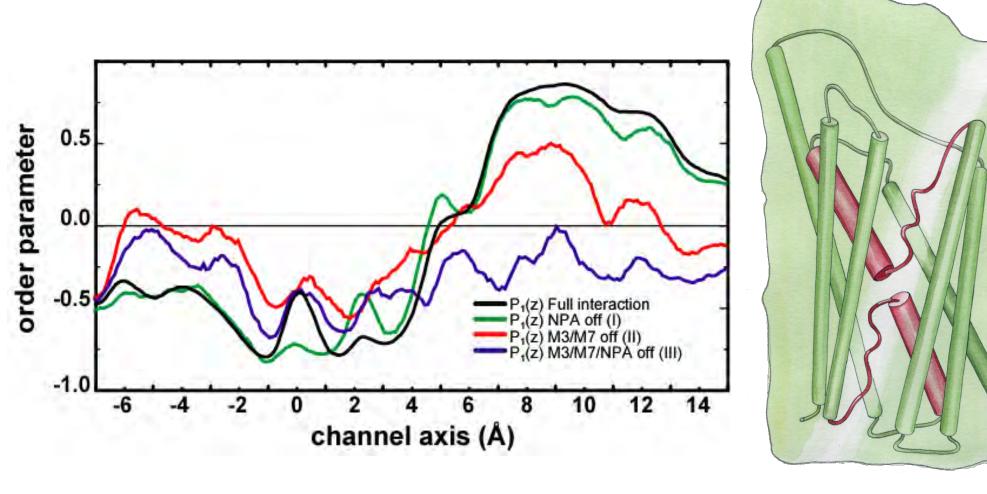


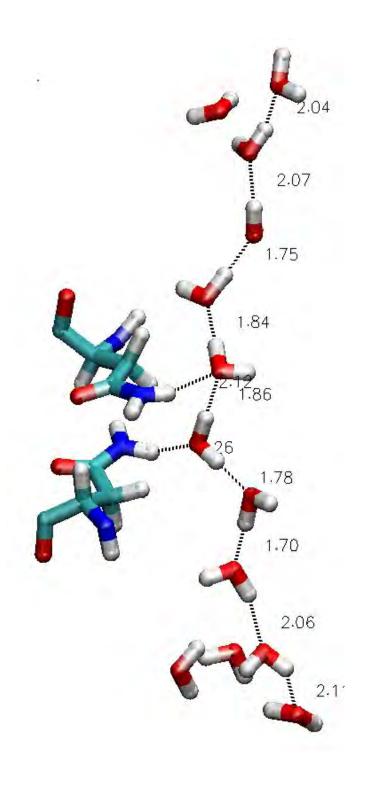
REMEMBER:

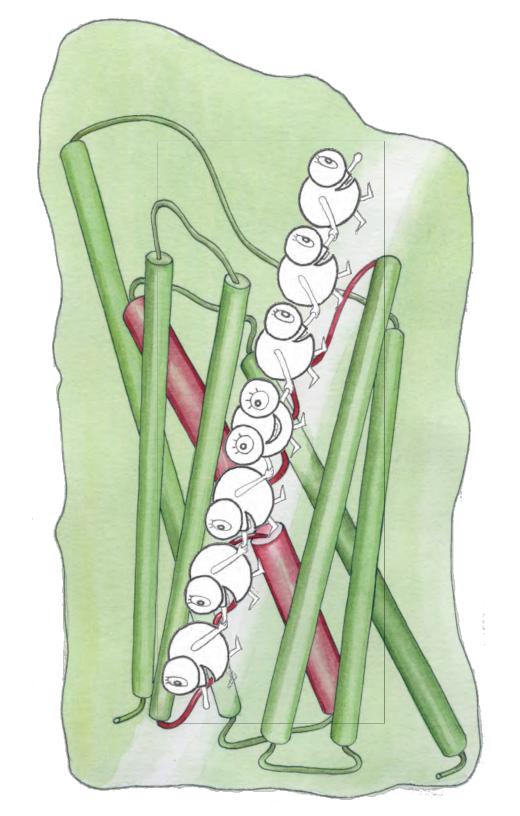
One of the most useful advantages of simulations over experiments is that you can modify the system as you wish: You can do modifications that are not even possible at all in reality!

This is a powerful technique to test hypotheses developed during your simulations. Use it!

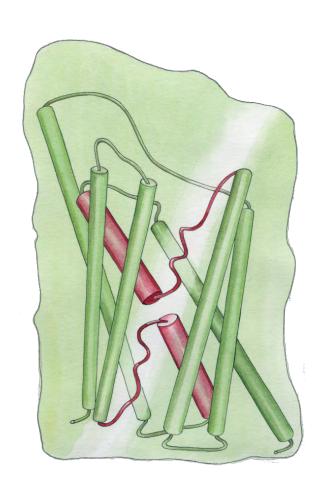
Electrostatic Stabilization of Water Bipolar Arrangement



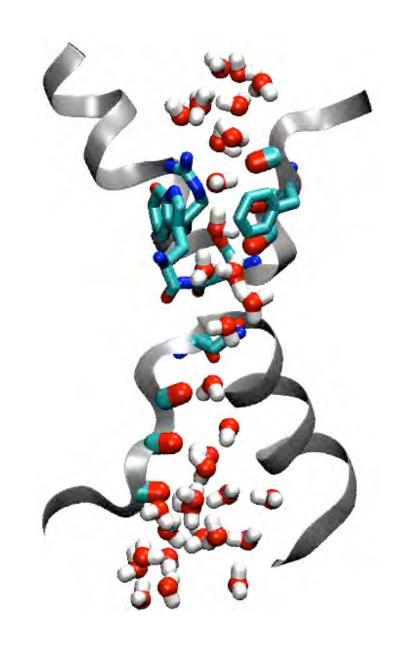




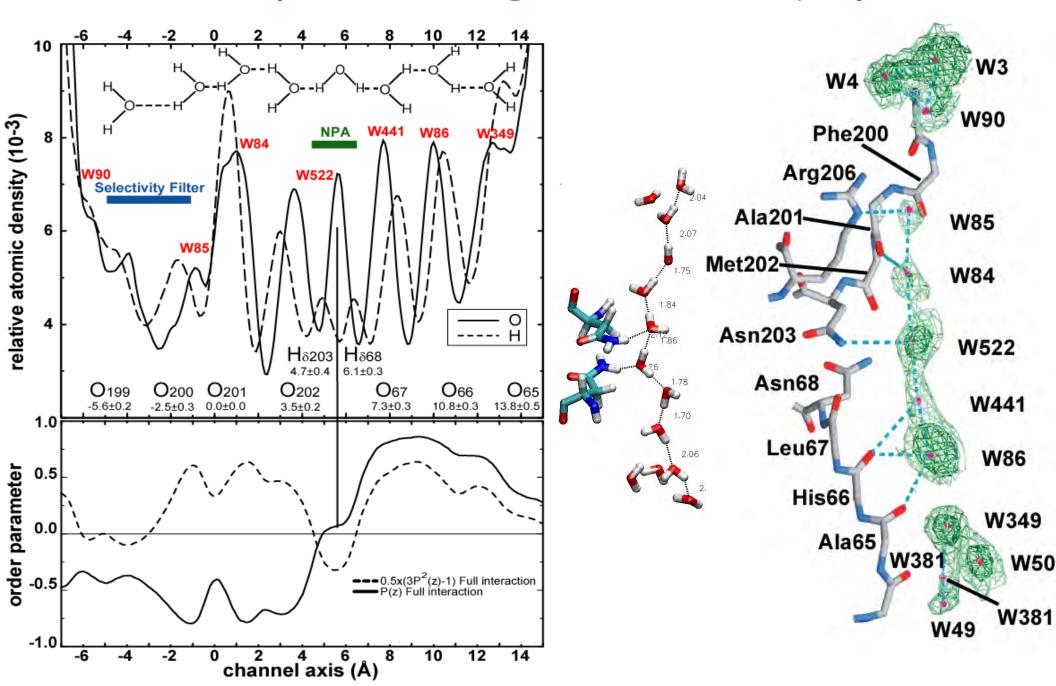
Characterizing Protein Forces



QM/MM MD of proton behavior in the channel



Water Bipolar Configuration in Aquaporins



Proton transfer through water

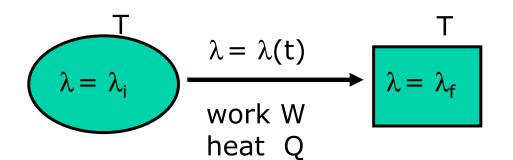
Battling the Timescale - Case I

Steered Molecular Dynamics is a non-equilibrium method by nature

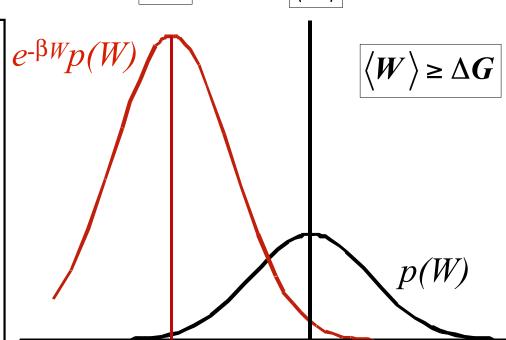
- A wide variety of events that are inaccessible to conventional molecular dynamics simulations can be probed.
- The system will be driven, however, away from equilibrium, resulting in problems in describing the energy landscape associated with the event of interest.

Jarzynski's Equality

Transition between two equilibrium states



$$\Delta \boldsymbol{G} = \boldsymbol{G}_f - \boldsymbol{G}_i$$



 ΔG

C. Jarzynski, *Phys. Rev. Lett.*, **78**, 2690 (1997)C. Jarzynski, *Phys. Rev. E*, **56**, 5018 (1997)

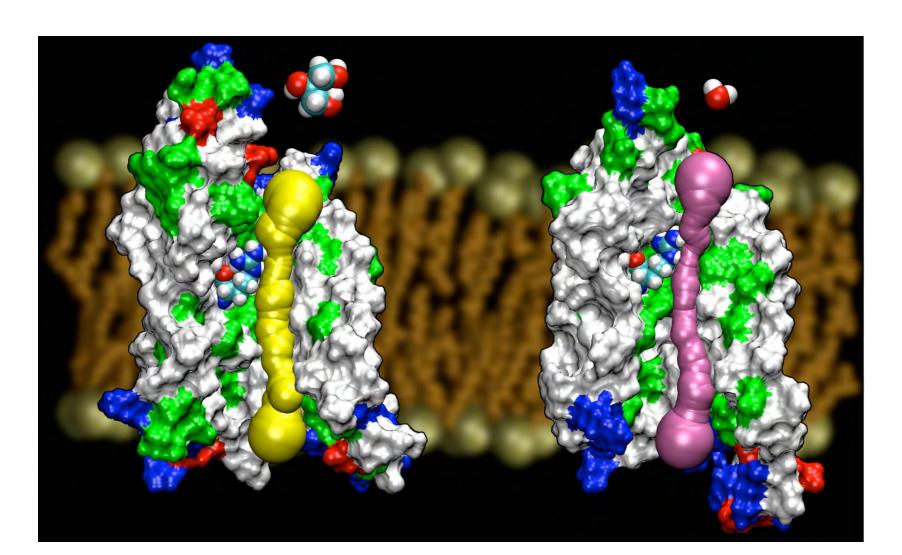
$$\langle e^{-\beta W} \rangle = e^{-\beta \Delta G}$$

In principle, it is possible to obtain free energy surfaces from <u>repeated</u> non-equilibrium experiments.

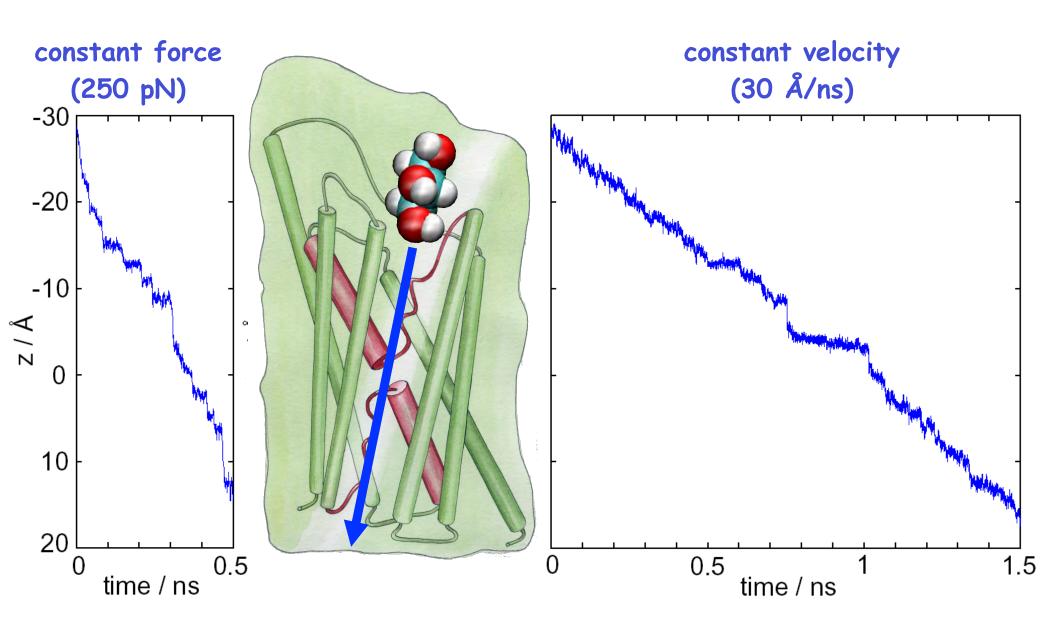
$$\beta = \frac{1}{k_B T}$$

AqpZ vs. GlpF

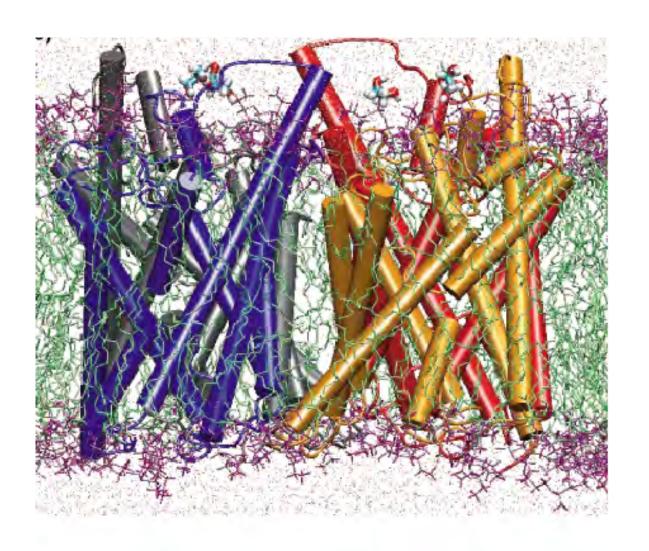
- Both from *E. coli*
- AqpZ is a pure water channel
- GlpF is a glycerol channel
- We have high resolution structures for both channels

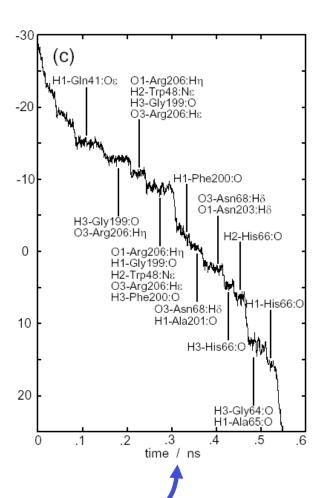


Steered Molecular Dynamics



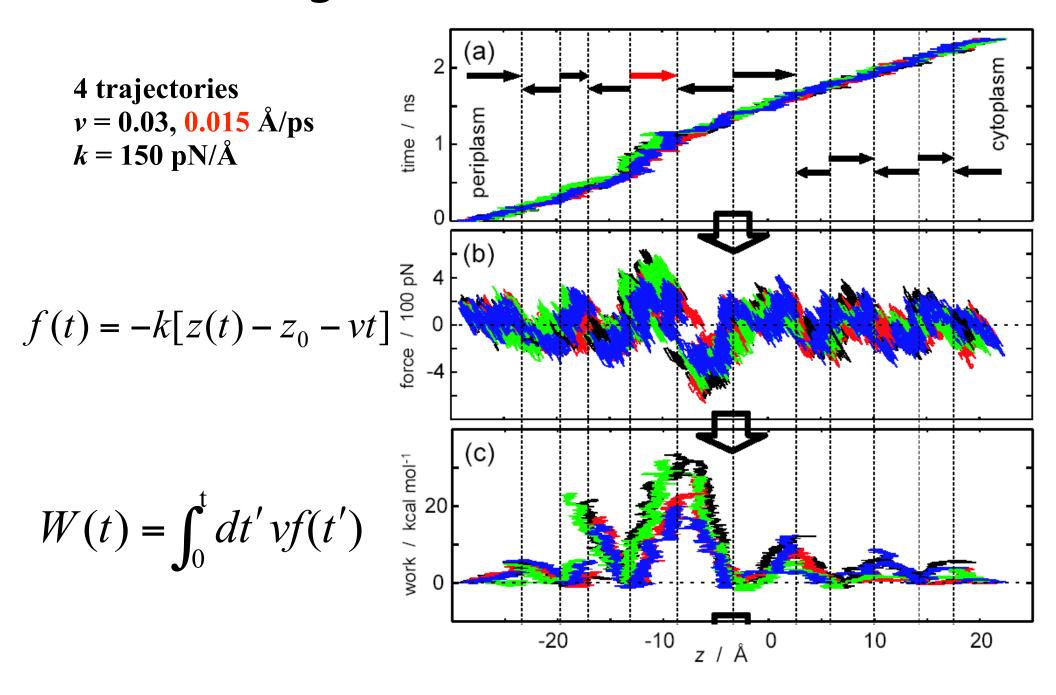
SMD Simulation of Glycerol Passage



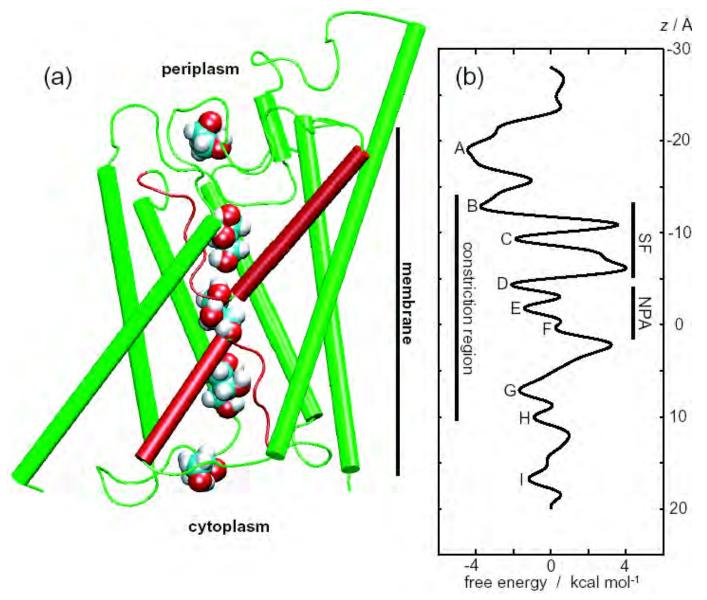


Trajectory of glycerol pulled by constant force

Constructing the Potential of Mean Force

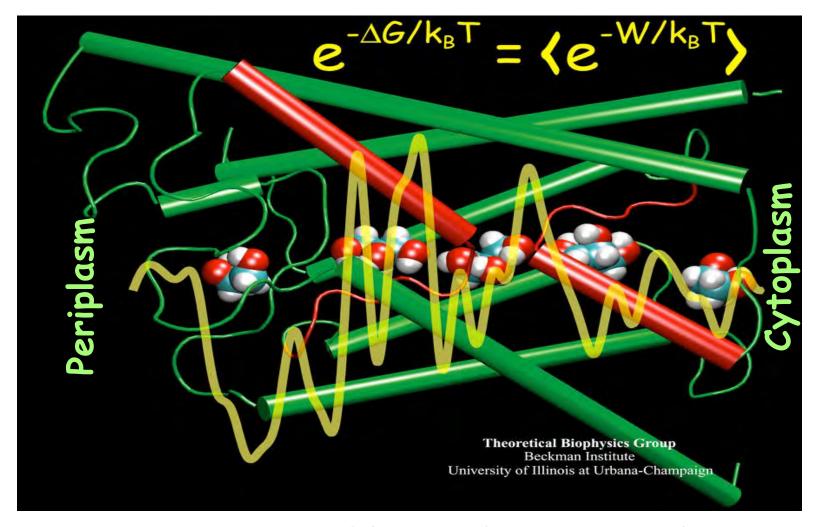


Features of the Potential of Mean Force



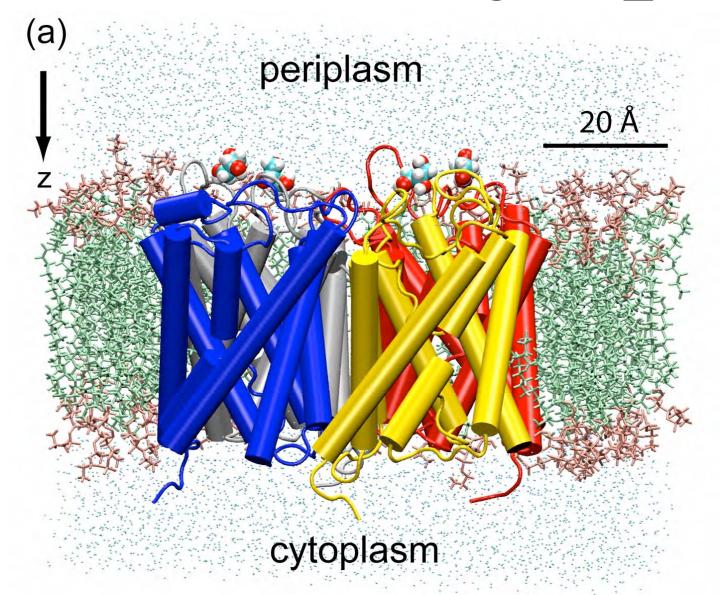
- · Captures major features of the channel
- The largest barrier \approx 7.3 kcal/mol; exp.: 9.6 \pm 1.5 kcal/mol Jensen et al., *PNAS*, 99:6731-6736, 2002.

Features of the Potential of Mean Force



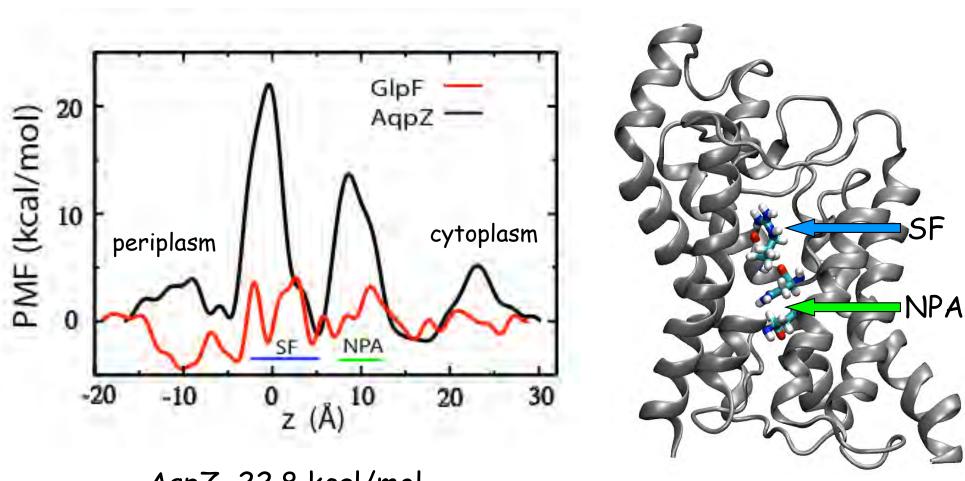
Asymmetric Profile in the Vestibules

Artificial induction of glycerol conduction through AqpZ



Y. Wang, K. Schulten, and E. Tajkhorshid Structure 13, 1107 (2005)

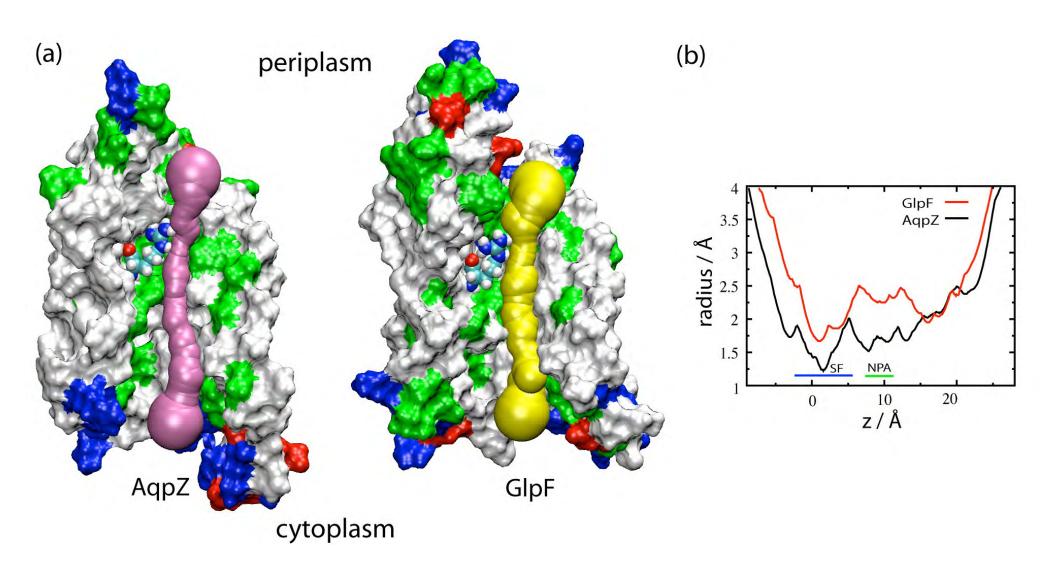
Three fold higher barriers



AqpZ 22.8 kcal/mol GlpF 7.3 kcal/mol

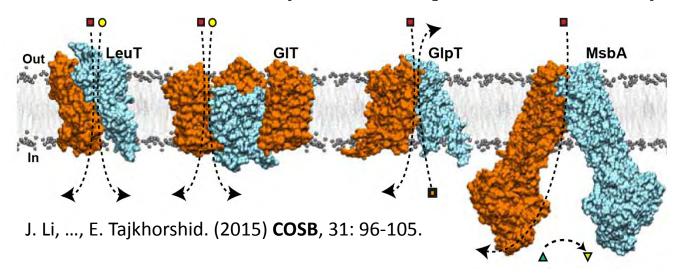
Y. Wang, K. Schulten, and E. Tajkhorshid *Structure* 13, 1107 *(*2005)

Could it be simply the size?



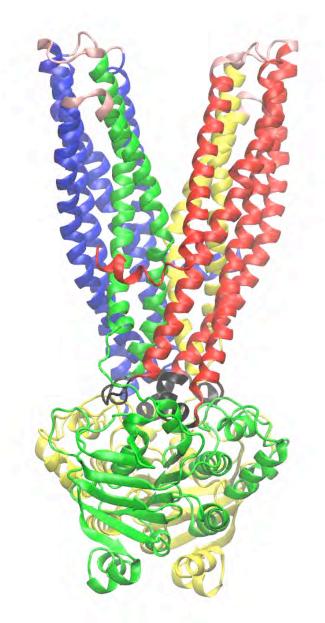
Y. Wang, K. Schulten, and E. Tajkhorshid *Structure* 13, 1107 (2005)

Battling the Timescale - Case II Biased (nonequilibrium) simulations



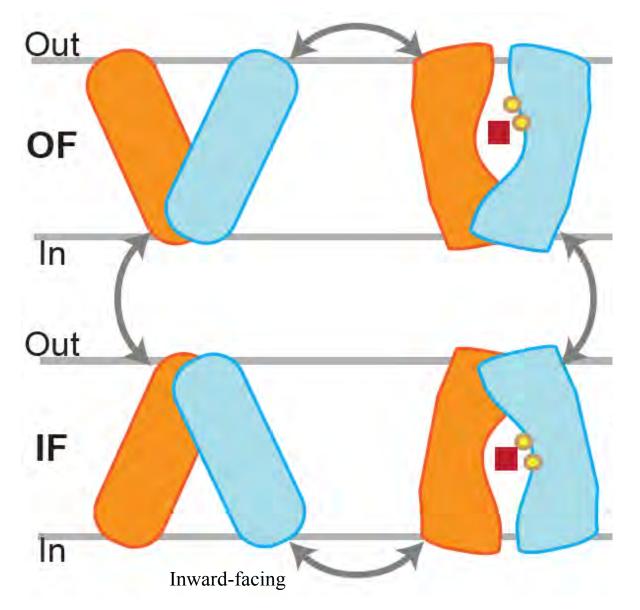
◆ Neurotransmitter Uptake

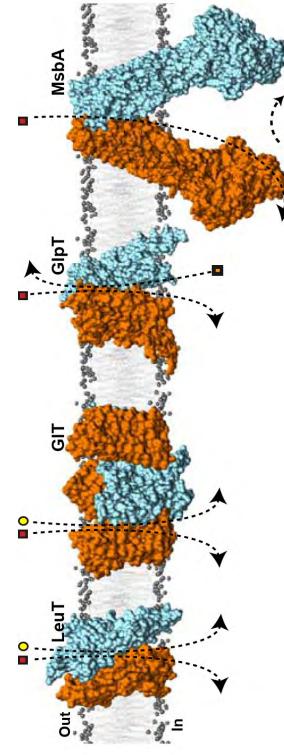
- » Norepinephrine, serotonin, dopamine, glutamate,...
- **♦** Gastrointestinal Tract
 - » Active absorption of nutrients
 - » Secretion of ions
- Kidneys
 - » Reabsorption
 - » Secretion
- **♦** Pharmacokinetics of all drugs
 - » Absorption, distribution, elimination
 - » Multi-drug resistance in cancer cells



Alternating Access Mechanism

Outward-facing

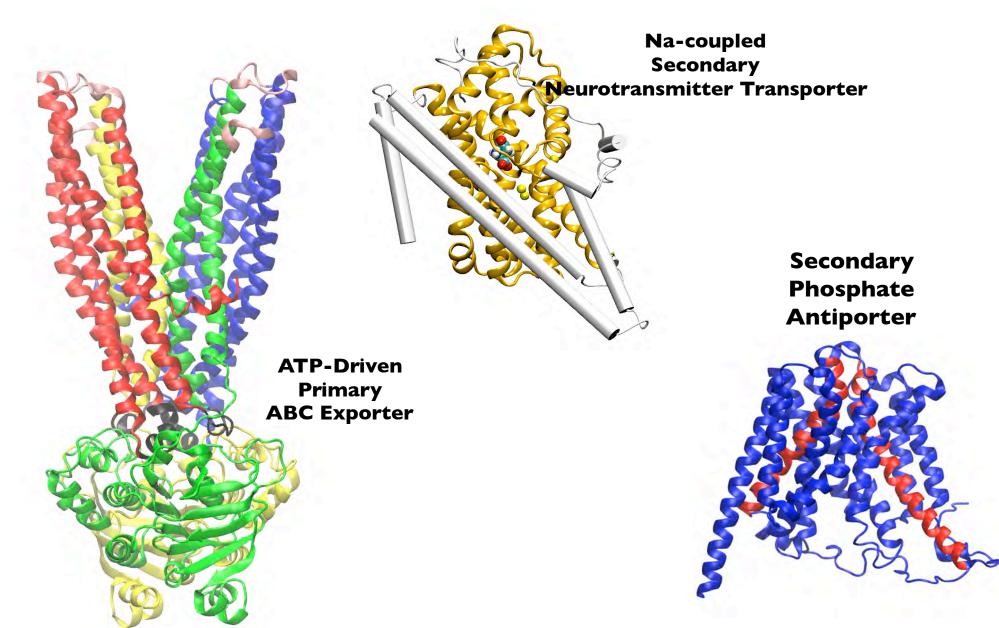




Jardetzky O. *Nature* **211**: 969–970 (1966)

J. Li, ..., E. Tajkhorshid. (2015) **COSB**, 31: 96-105.

Diverse Structural Transitions Involved



Non-equilibrium methods are required.

Complex Processes Require Complex Treatments

I.1 Defining Practical Collective Variables

Empirical search for practical collective variables for inducing the conformational changes involved in the transition.

I.2 Optimizing the Biasing Protocols

Systematic search for a practical biasing protocol by using different combinations of collective variables.

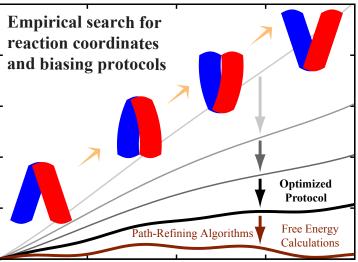


Mahmoud Moradi

II. Optimizing the Transition Pathway

Use all of the conformations available to generate the most reliable transition pathway:

- 1. Bayesian approach for combining the data
- 2. Post-hoc string method (analysis tool)
- 3. String method with swarms of trajectories



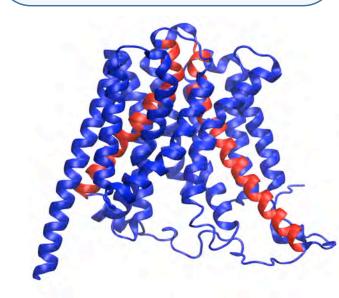
Reaction Coordinate

III.1 Free Energy Calculations

Using the most relevant collective variables (from I.1), biasing protocol (from I.2), and initial conformations (from I.2).

III.2 Assessing the Sampling Efficiency

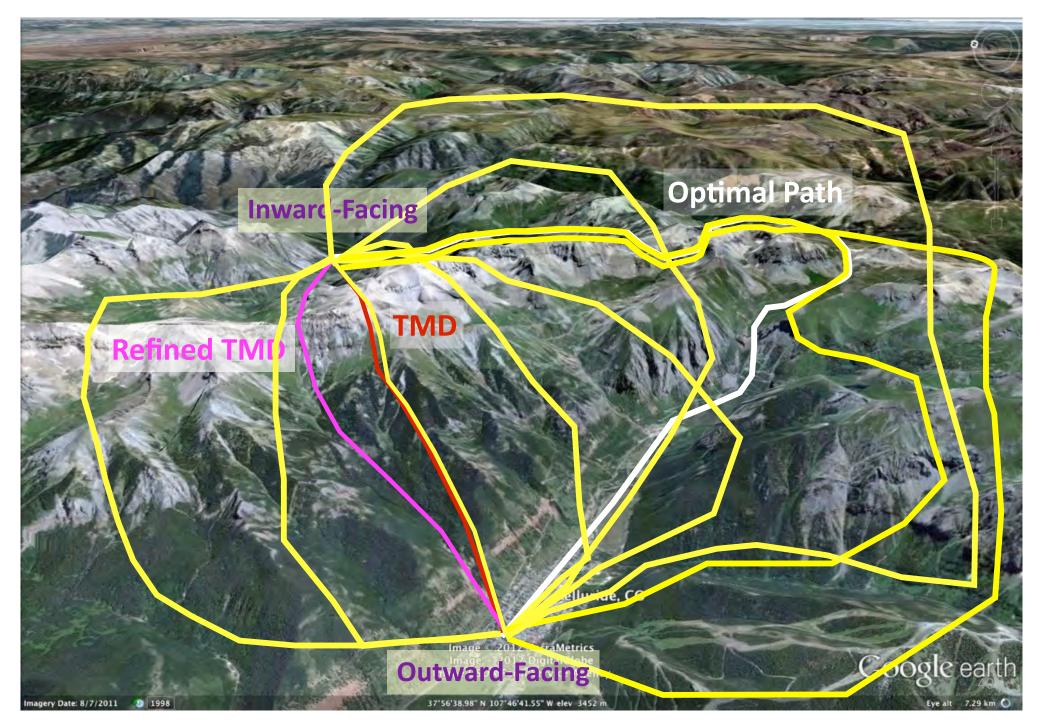
Detecting the poorly sampled, but potentially important regions, e.g., by using PCA.



- M. Moradi and ET (2013) **PNAS**, 110:18916–18921.
- M. Moradi and ET (2014) JCTC, 10: 2866-2880.
- M. Moradi, G. Enkavi, and ET (2015) Nature Comm., 6:8393.

Work

Aggressive Search of the Space



Non-equilibrium Driven Molecular Dynamics:

Applying a time-dependent external force to induce the transition

Along various pathways/mechanisms (collective variables)

Harmonic constant

Initial state

$$U_{dr}(\mathbf{x},t) = \frac{1}{2}k \left(\boldsymbol{\xi}(\mathbf{x}) - \boldsymbol{\xi}_A^{\dagger} + (\boldsymbol{\xi}_B - \boldsymbol{\xi}_A) \frac{t}{T} \right)^2$$
Final state

Biasing potential

Collective variables:

RMSD, distance,

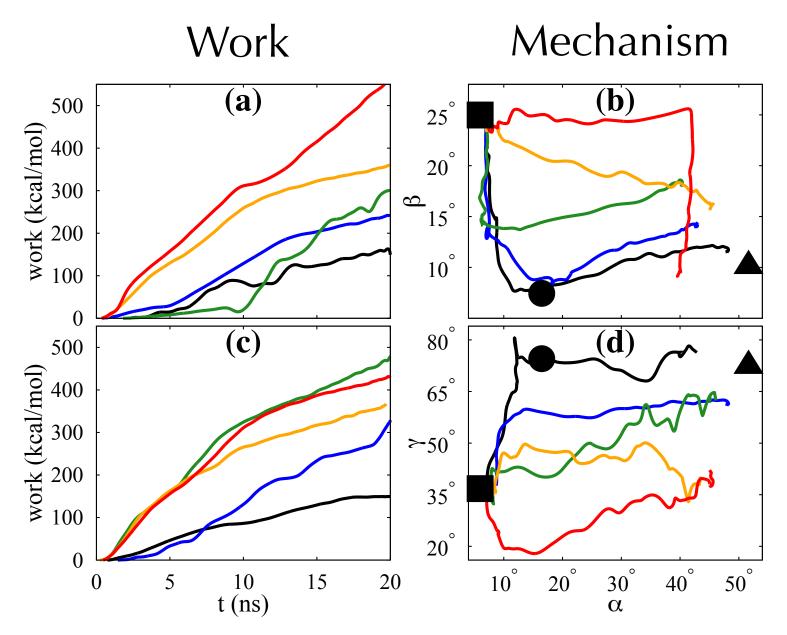
R_g, angle, ...

orientation quaternion

Total simulation time

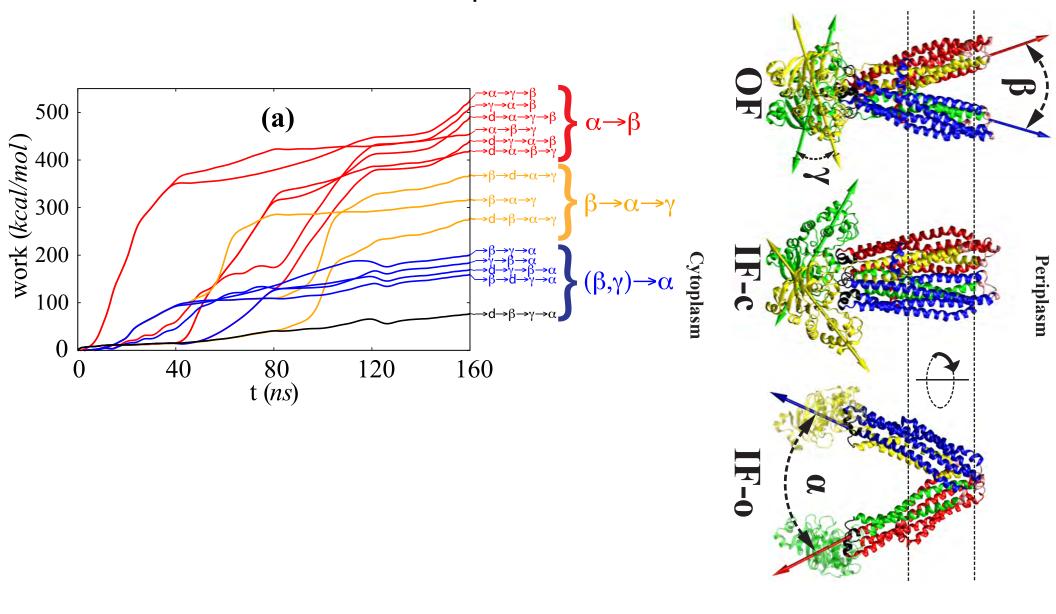
- M. Moradi and ET (2013) **PNAS**, 110:18916–18921.
- M. Moradi and ET (2014) JCTC, 10: 2866-2880.
- M. Moradi, G. Enkavi, and ET (2015) Nature Comm., 6:8393.

Progressively Optimizing the Biasing Protocol/Collective Variable using non-Equilibrium Work as a Measure of the Path Quality



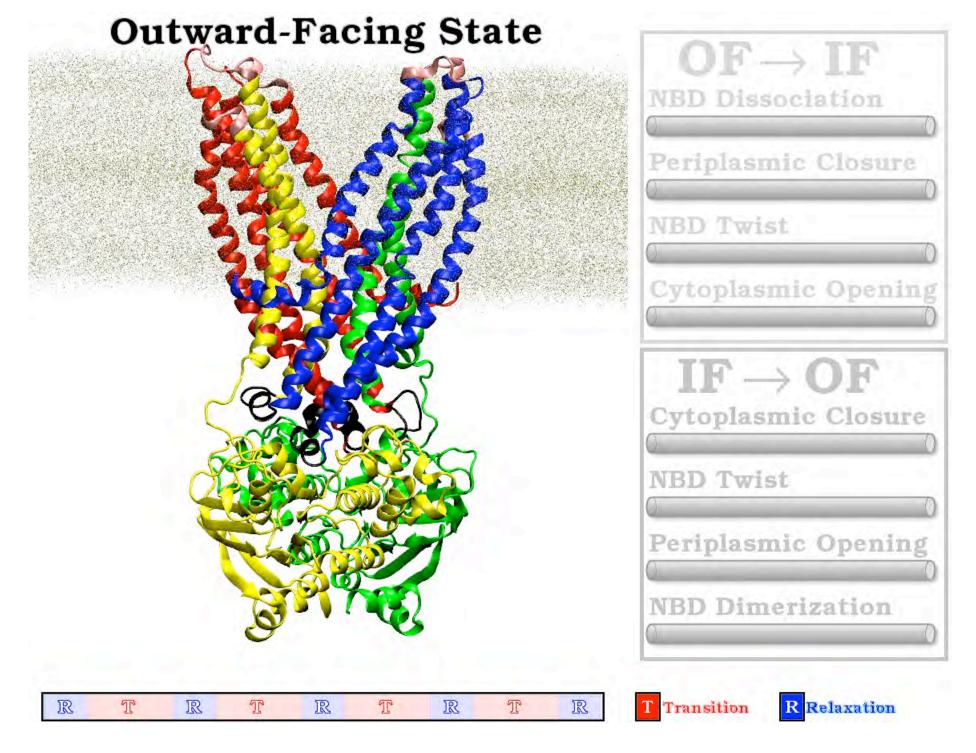
Example set taken from a subset of 20 ns biased simulations

Mechanistic Insight From Transition Pathways in ABC exporters from Non-Equilibrium Simulations

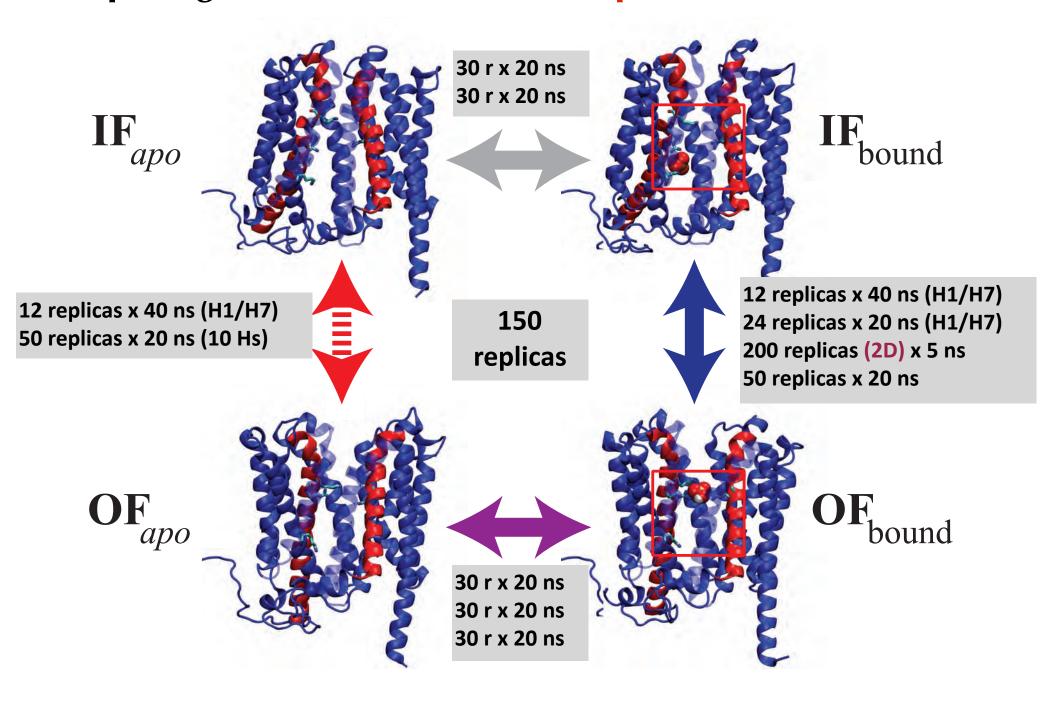


M. Moradi and ET (2013) PNAS, 110:18916–18921.

M. Moradi and ET (2014) JCTC, 10: 2866–2880.



Describing a Complete Cycle (Adding Substrate) Requiring a Combination of Multiple Collective Variables



Simulation protocols

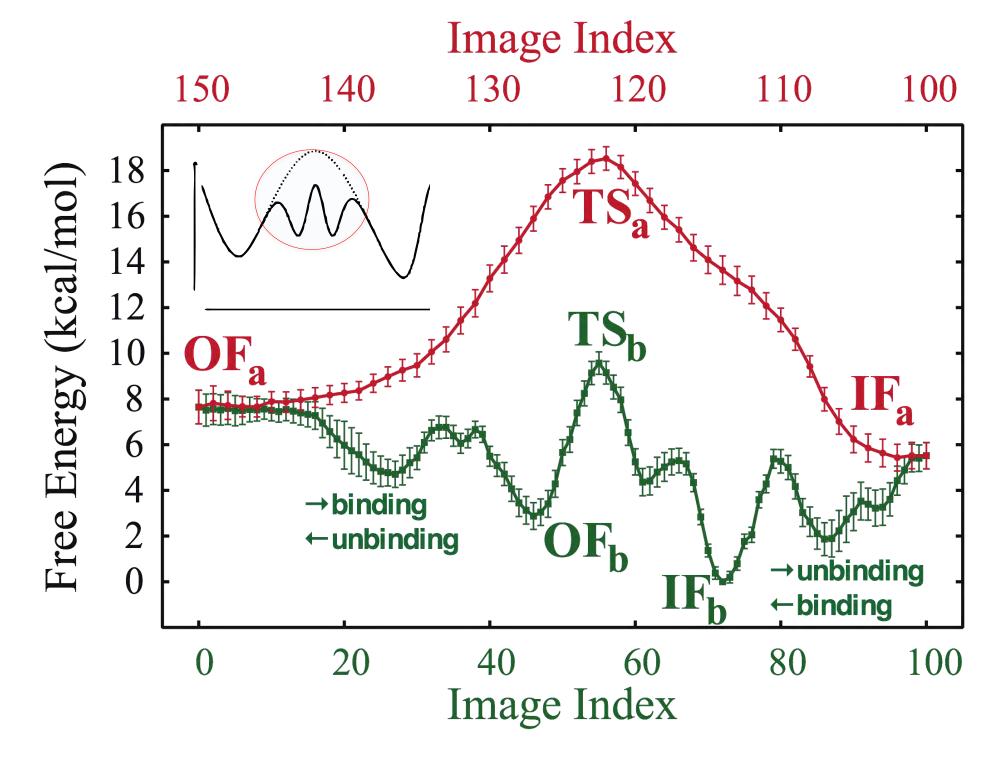
		Transition	Technique	Collective Variables	# of Replicas × Runtime		
	1		BEUS	(Q_1,Q_7)	$12 \times 40 \text{ ns}$	=	0.5 μs
	2	$IF_a \Leftrightarrow OF_a$	SMwST	{Q}	1000×1 ns	=	1 μs
	3		BEUS	{Q}	$50 \times 20 \text{ ns}$	=	1 μs
	4	$-IF_a \Leftrightarrow IF_b$	BEUS	Z_{Pi}	$30 \times 40 \text{ ns}$	=	1.2 μs
	5		BEUS	$(\{Q\},Z_{Pi})$	$30 \times 40 \text{ ns}$	=	1.2 µs
	6	$OF_a \Leftrightarrow OF_b$	BEUS	Z_{Pi}	$30 \times 40 \text{ ns}$	=	1.2 μs
	7		BEUS	$(\{Q\},Z_{Pi})$	$30 \times 40 \text{ ns}$	=	1.2 µs
	8		BEUS	(Q_1,Q_7)	$24 \times 20 \text{ ns}$	=	0.5 μs
	9	$IF_b \Leftrightarrow OF_b$	BEUS	Z_{Pi}	$15 \times 30 \text{ ns}$	=	0.5 μs
	10		2D BEUS	$(\Delta RMSD, Z_{Pi})$	200×5 ns	=	1 μs
	11		SMwST	$(\{Q\},Z_{Pi})$	1000×1 ns	=	1 μs
	12		BEUS	$(\{Q\},Z_{Pi})$	$50 \times 20 \text{ ns}$	=	1 μs
ı	13	Full Cycle	BEUS	$(\{Q\},Z_{Pi})$	150 × 50 ns	=	7.5 μs
	Total Simulation Time 18.7 μs						
$\begin{array}{cccccccccccccccccccccccccccccccccccc$							
		BEUS → PHSM					





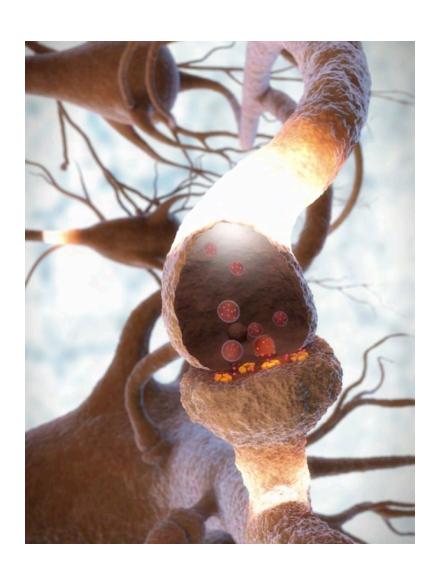
---> Nonequilibrium

SMwST

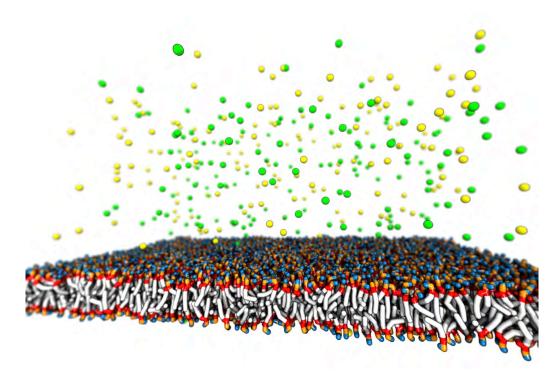


M. Moradi, G. Enkavi, and ET (2015) Nature Communication, 6: 8393.

Battling the Timescale - Case III Multiscale Simulations



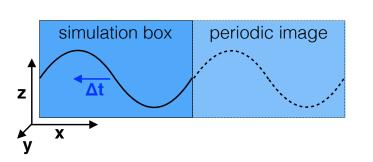
Membrane Budding/Fusion

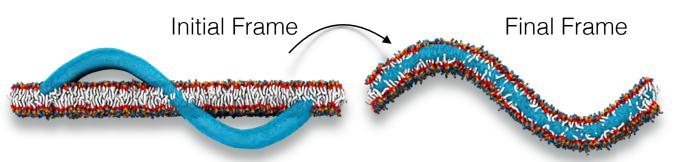


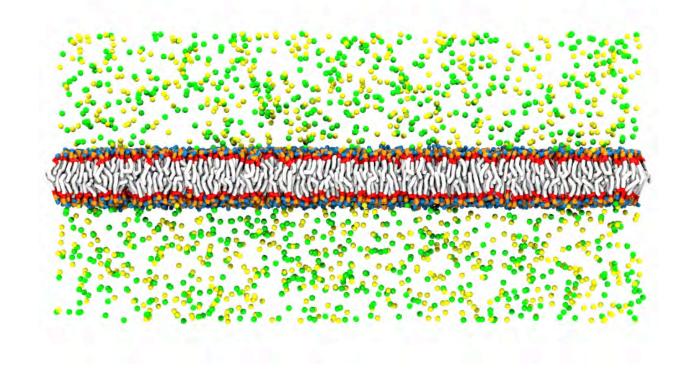
Combining multiple replica simulations and coarsegrained models to describe membrane fusion

Workflow for Multi-Scale Modeling

Parametrically Defined Sine Function

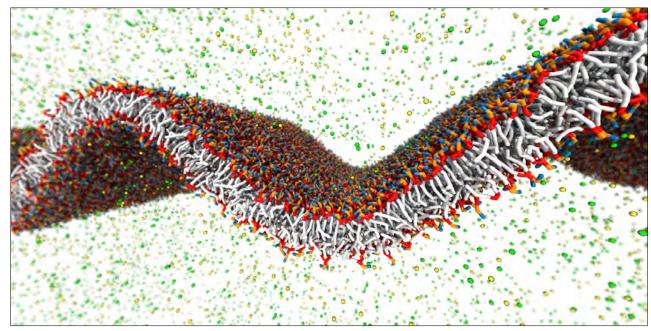




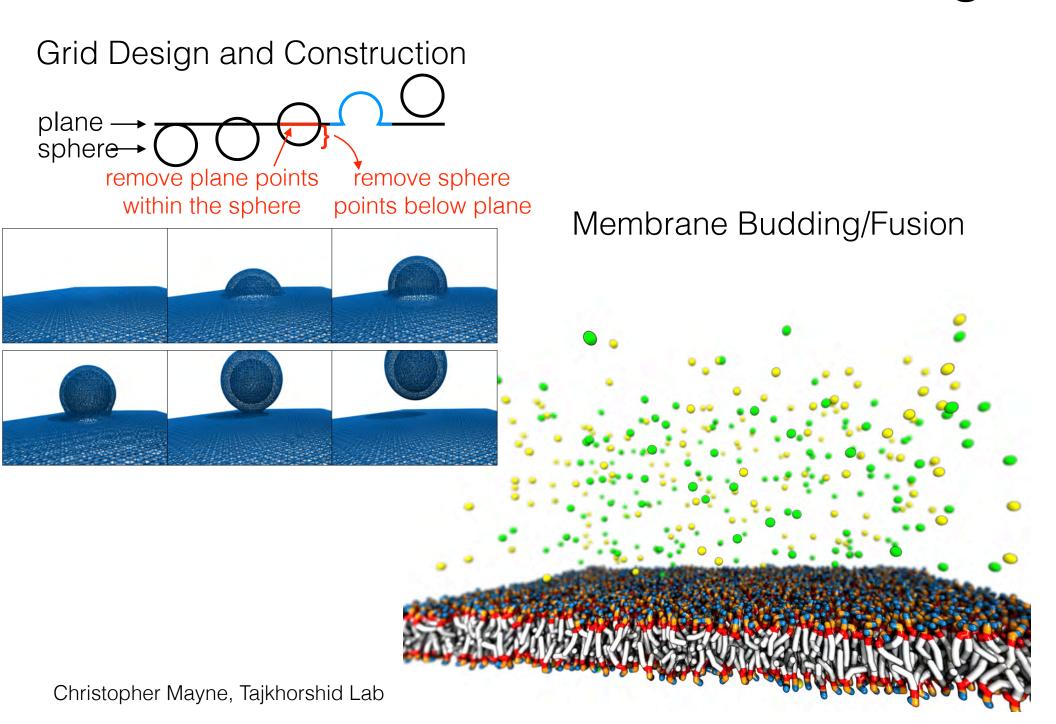


Workflow for Multi-Scale Modeling



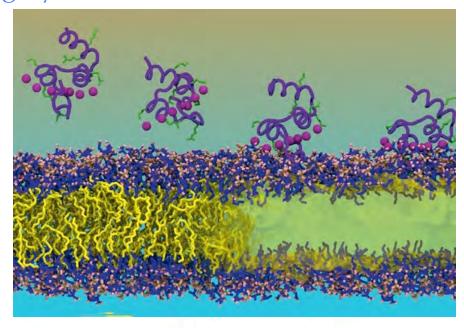


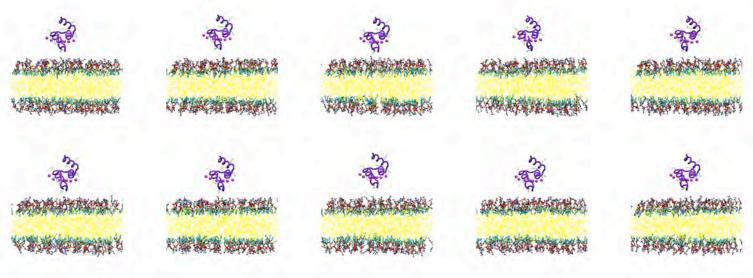
Workflow for Multi-Scale Modeling



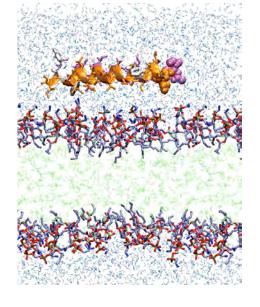
Battling the Timescale - Case IV Reduced Representations

Highly Mobile Membrane Mimetic model





GpA insertion in 12 ns



Specific lipids regulate various functional aspects of membrane proteins

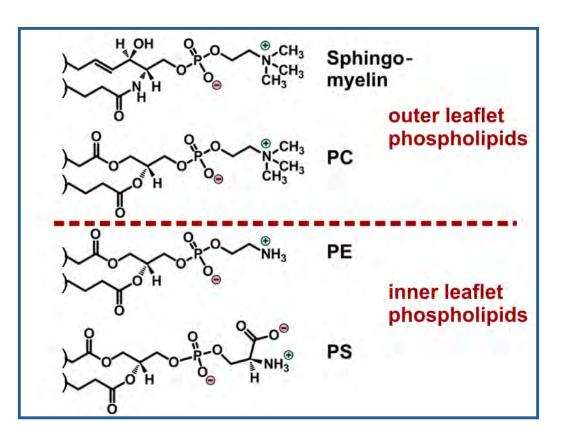
Integral membrane proteins

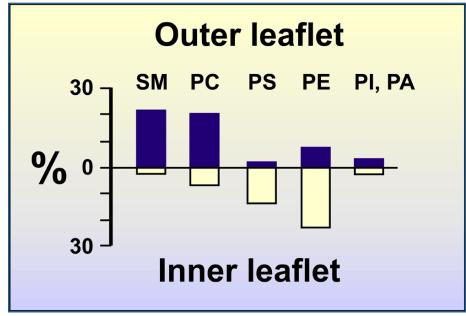
Peripheral membrane proteins

Lipid-Dependent Regulation and Activity of Peripheral Membrane Proteins

- ◆ Membrane binding is a key regulatory step in the function of diverse proteins:
 - ◆ Cytoplasmic enzymes (kinases, Ras, P450, synaptotagmin, ...)
 - ◆ Coagulation factors (GLA and C2 domains)
 - ◆ Membrane sculpting proteins (BAR domain)
 - ◆ Pathogenic systems viral fusion peptides, synuclein,
 - ◆ Immune/apoptotic system (TIM proteins)
- **→ Lipid-specificity** is a common feature:
 - ◆ Mostly at the level of **head groups**: PS, PG, PIP2, PA, ...
 - ◆ Requiring all-atom representation of the head groups
 - ◆ Slow lateral diffusion of lipids within a bilayer environment makes simulation studies of membrane-associated phenomena even more challenging

Lipid Dependent Binding and Activation

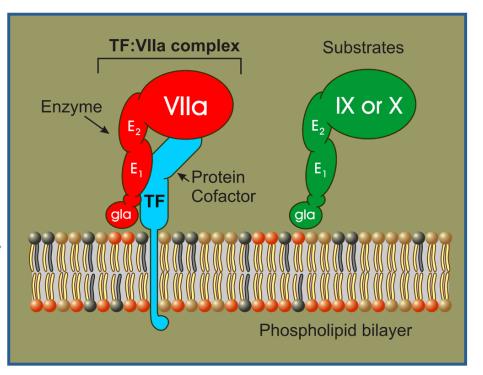




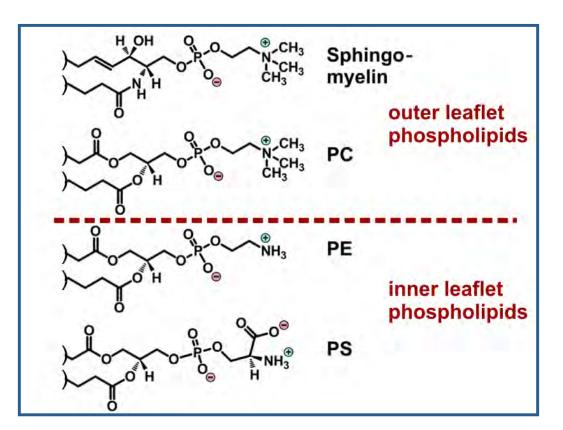
Affinity is controlled by lipid content

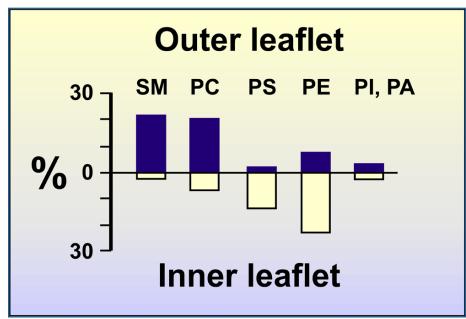
Leaflet asymmetry is vital for coagulation

Courtesy of Jim Morrissey, UIUC



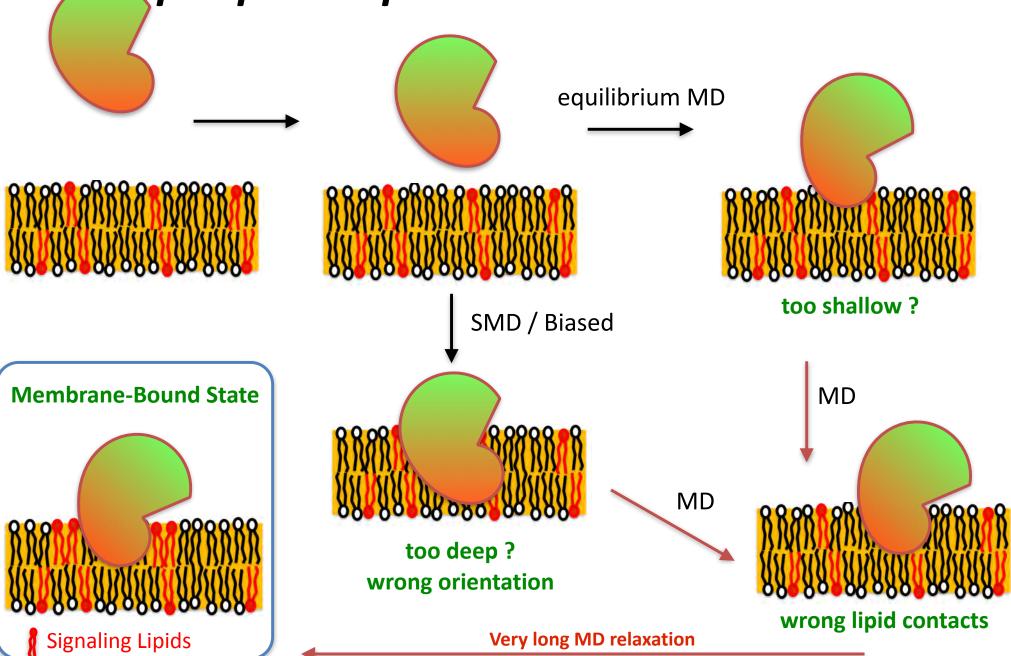
Lipid Dependent Binding and Activation



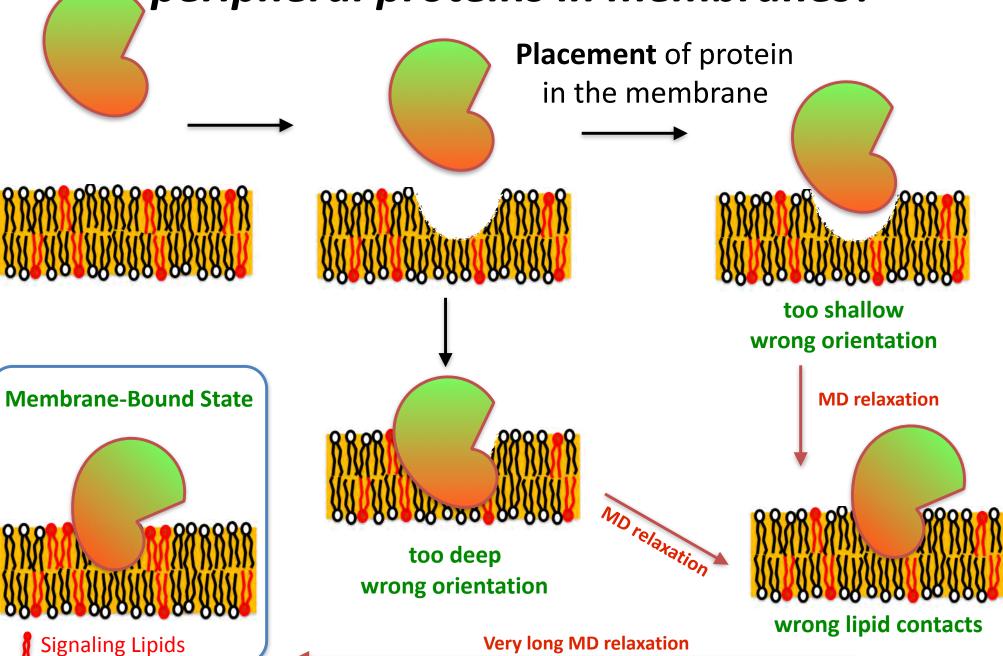


Mode and specificity of lipid-protein interactions constitute one of the main mechanistic aspects

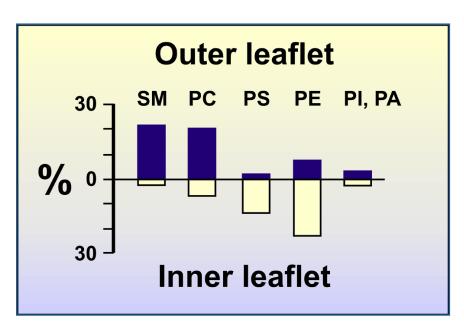
How do we construct an initial model for peripheral proteins in membranes?



How do we construct an initial model for peripheral proteins in membranes?

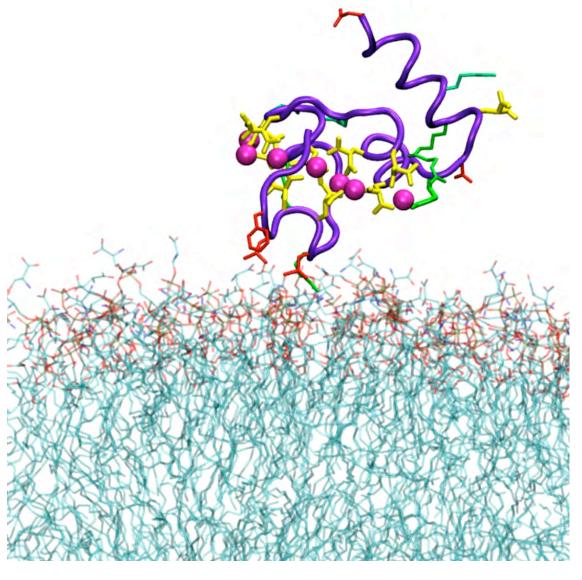


Simulation of Binding with Full Membrane Representation



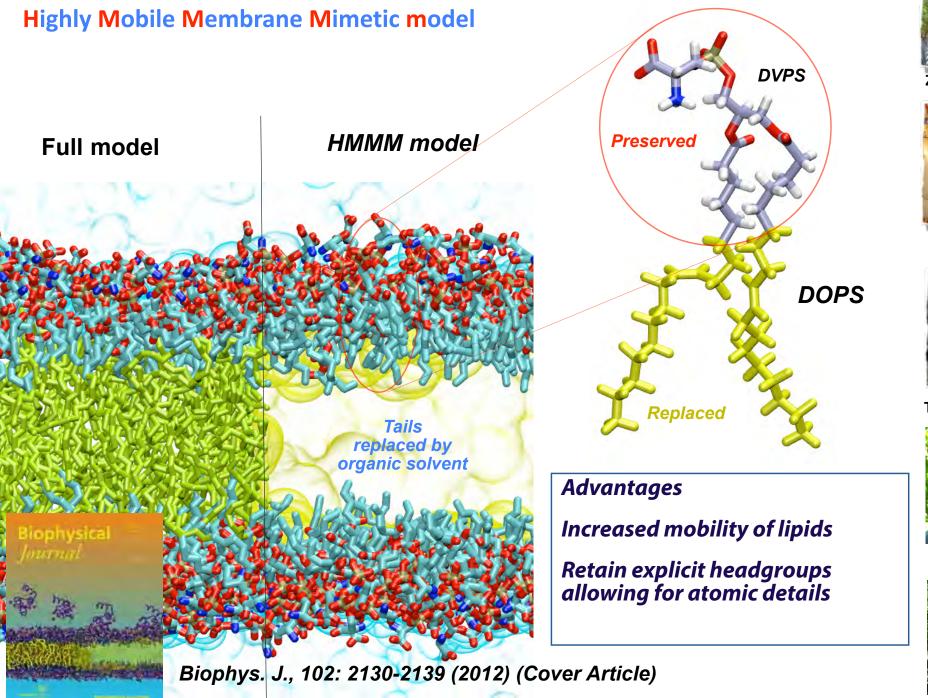
Partial list of technical problems:

- Biased simulations
- Unknown depth of insertion
- Single binding event
- Frequently failing
- Minimal lipid reorganization



Z. Ohkubo and E. T., **Structure**, 16: 72-81 (2008)

HMMM model





Zenmei Ohkubo



Mark Arcario



Taras Pogorelov



Josh Vermaas

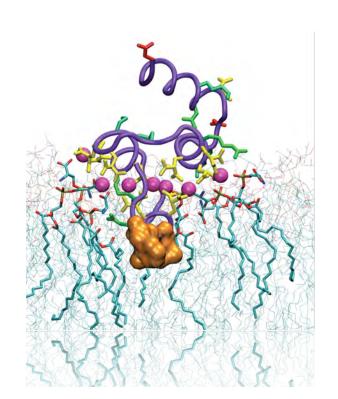


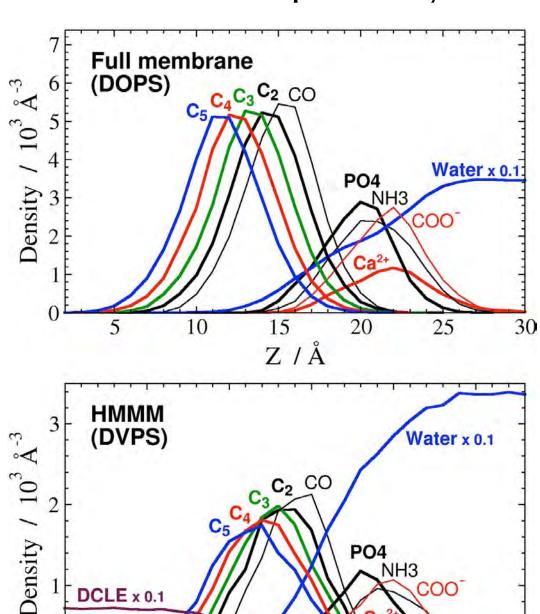
Javier Baylon

HMMM- Preserving the "Face" of the Lipid Bilayer

Perfect match in the membrane profile particularly in the head group region

Critical for proper description of lipid protein interactions





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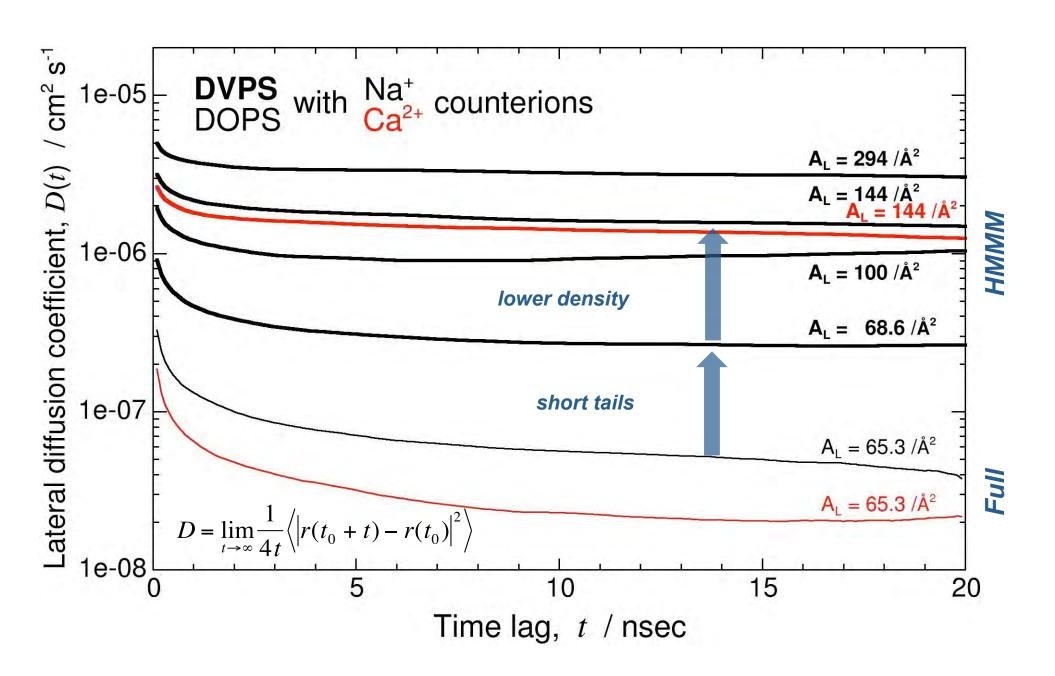
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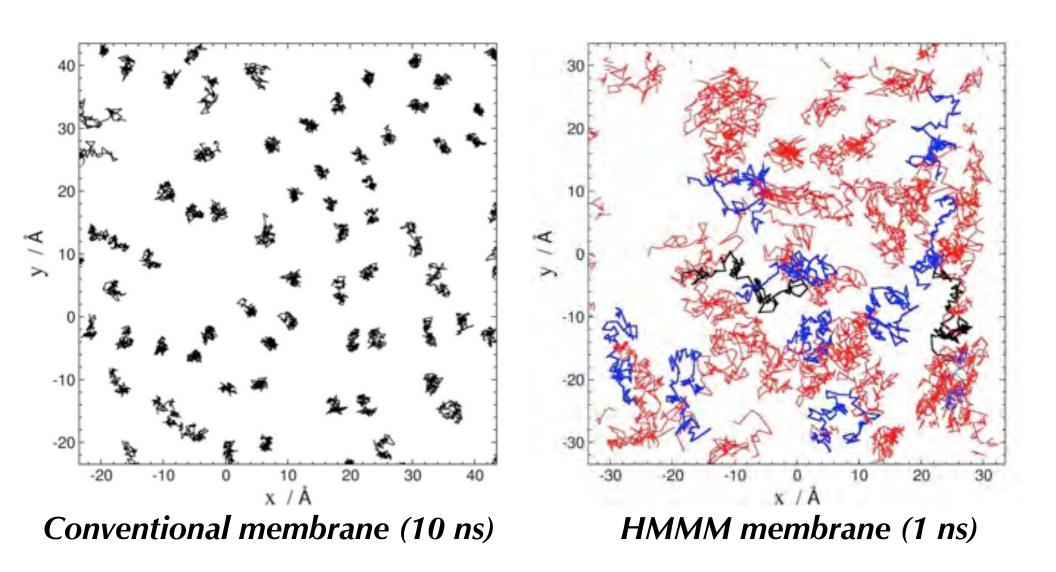
Enhanced Lipid Lateral Diffusion

Without Compromising Atomic Details of the Headgroups

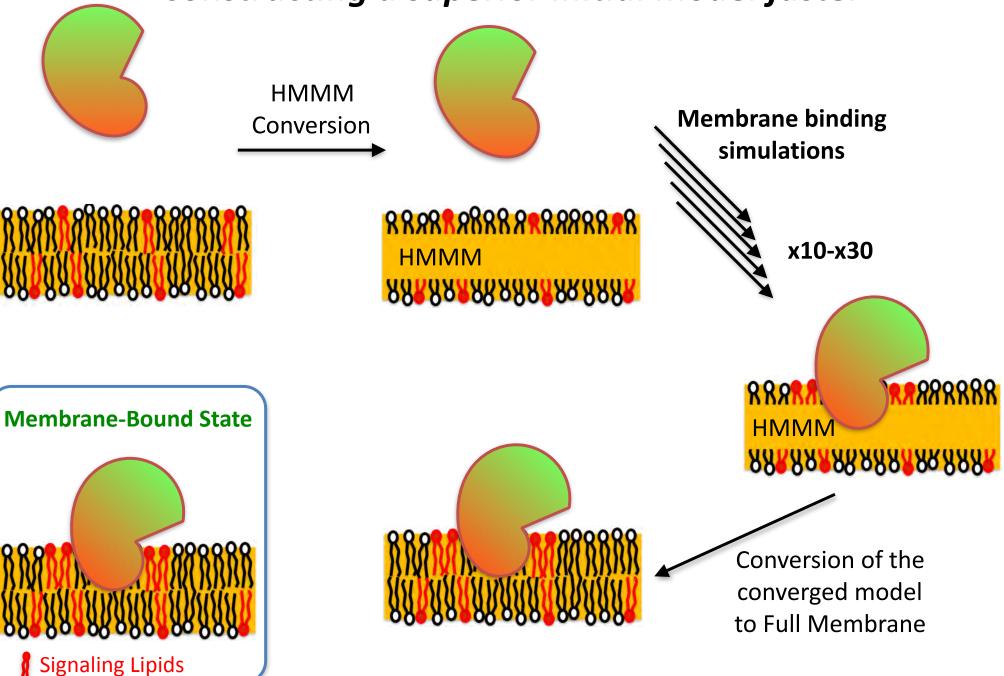


Enhanced Lipid Lateral Diffusion

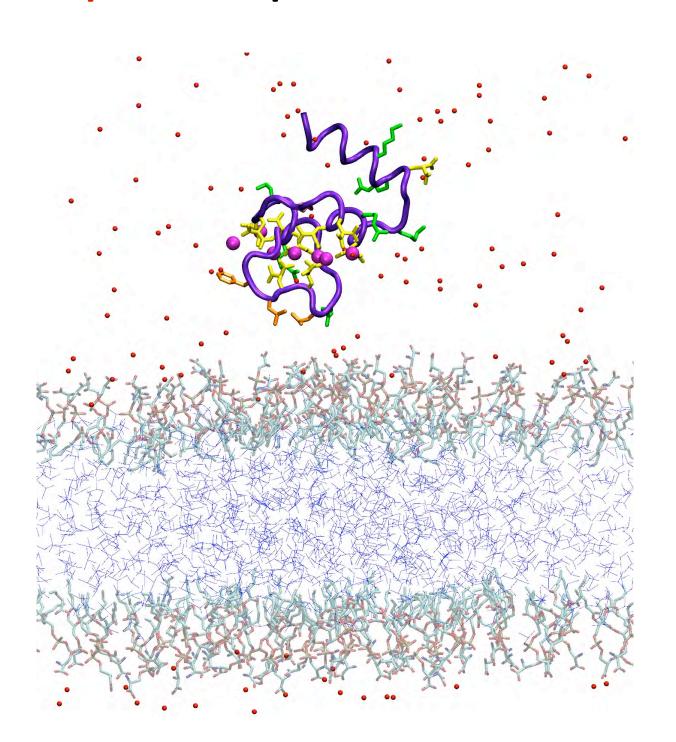
Without Compromising Atomic Details of the Headgroups



HMMM accelerated sampling of lipid-protein interactions Constructing a superior initial model faster



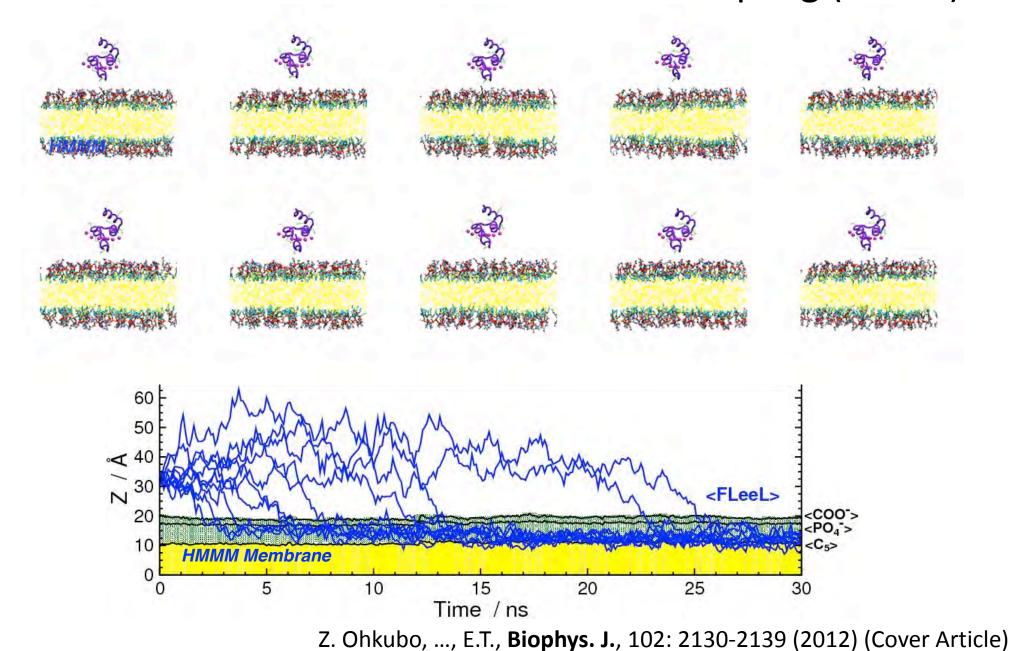
PS-Dependent Spontaneous Insertion of FVII-GLA





Zenmei Ohkubo

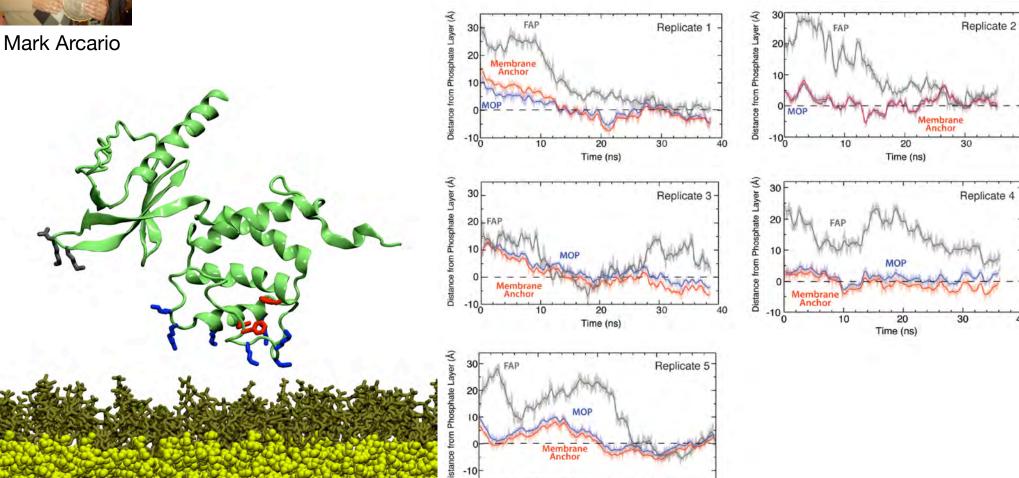
Spontaneous, Unbiased Membrane Binding Accelerated Process Allows for better sampling (n = 10)





PS-Dependent Membrane Binding of Talin

Five independent membrane binding simulations



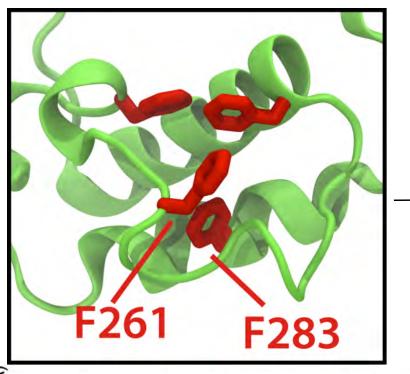
Final model converted to **full membrane**Stable in 100 ns simulations

M. Arcario and ET, **Biophys. J.**, 107: 2059–2069 (2014).

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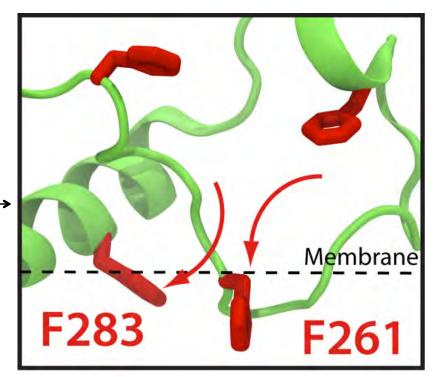
Time (ns)

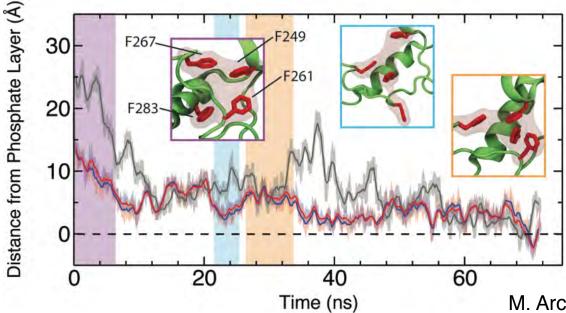
Revealing the *Hydrophobic Anchor*



20 ns

Membrane Binding Simulation

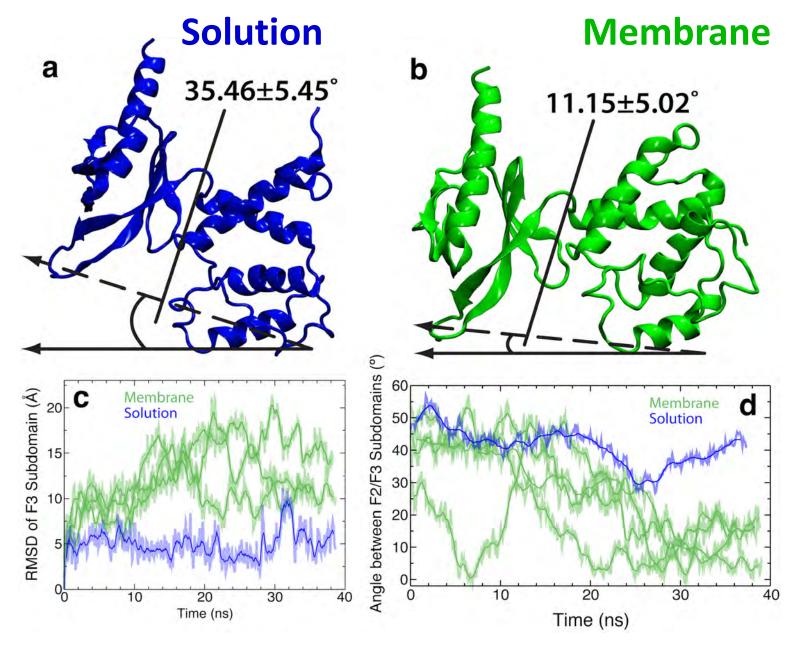




- Snorkeling of lysine acts as a switch which releases a conserved phenylalanine anchor (F261 & F283) into the membrane
- Reformation of the hydrophobic pocket causes looser binding of talin; suggests a mechanism for unbinding of protein

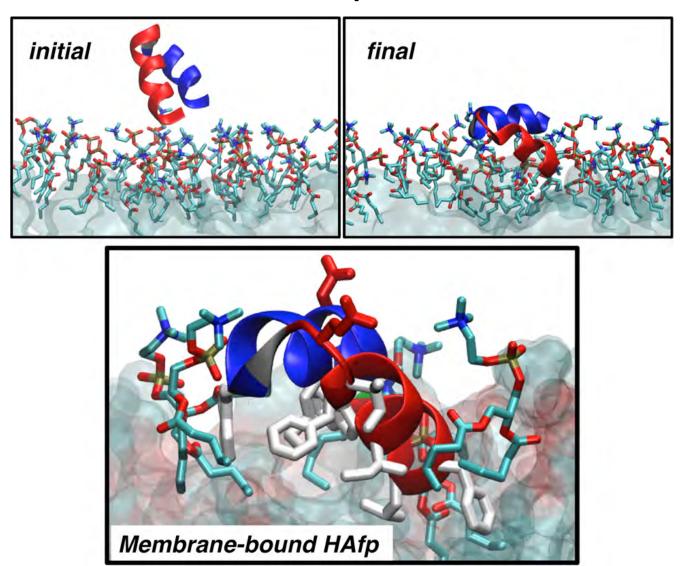
M. Arcario and ET, **Biophys. J.**, 107: 2059–2069 (2014).

Membrane Induced Domain Rearrangement of Talin



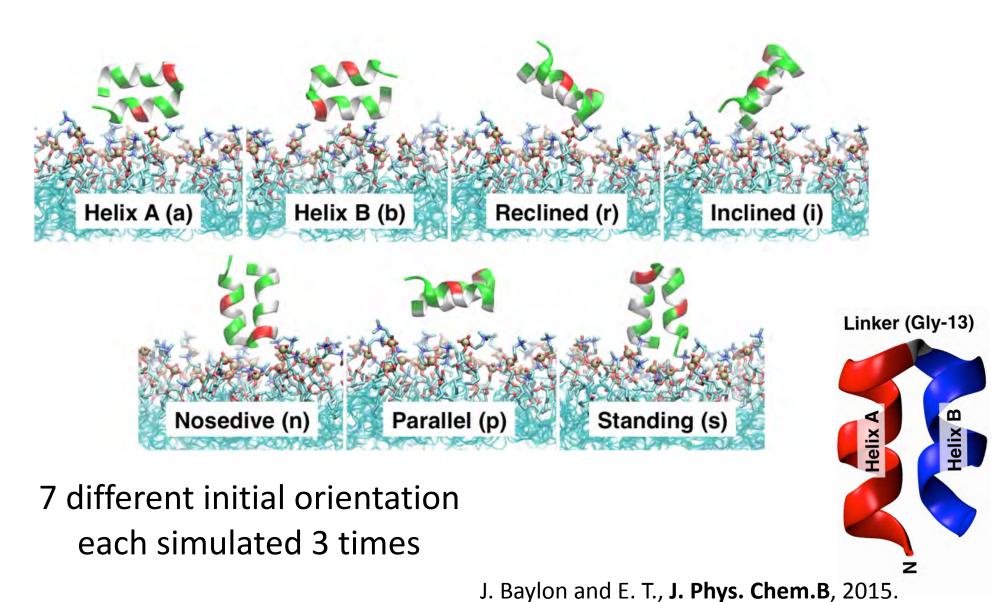
M. Arcario and ET, **Biophys. J.**, 107: 2059–2069 (2014).

Membrane Binding of Influenza Hemagglutinin Fusion Peptide



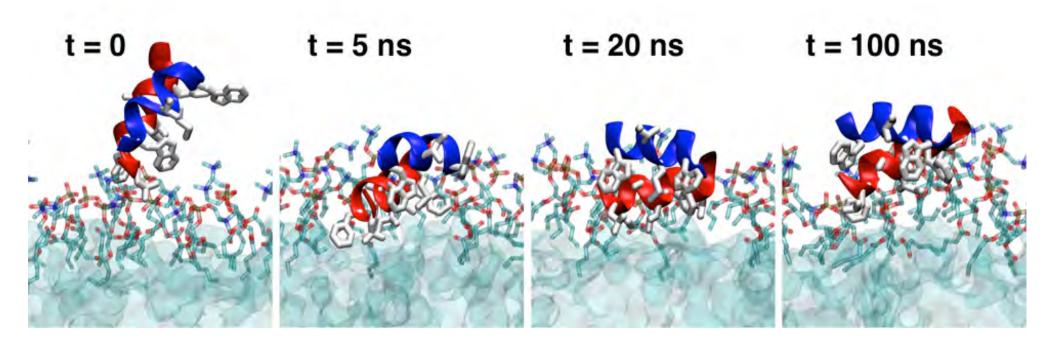
J. Baylon and E. T., J. Phys. Chem.B, 2015, in press.

Membrane Binding of Influenza Hemagglutinin Fusion Peptide



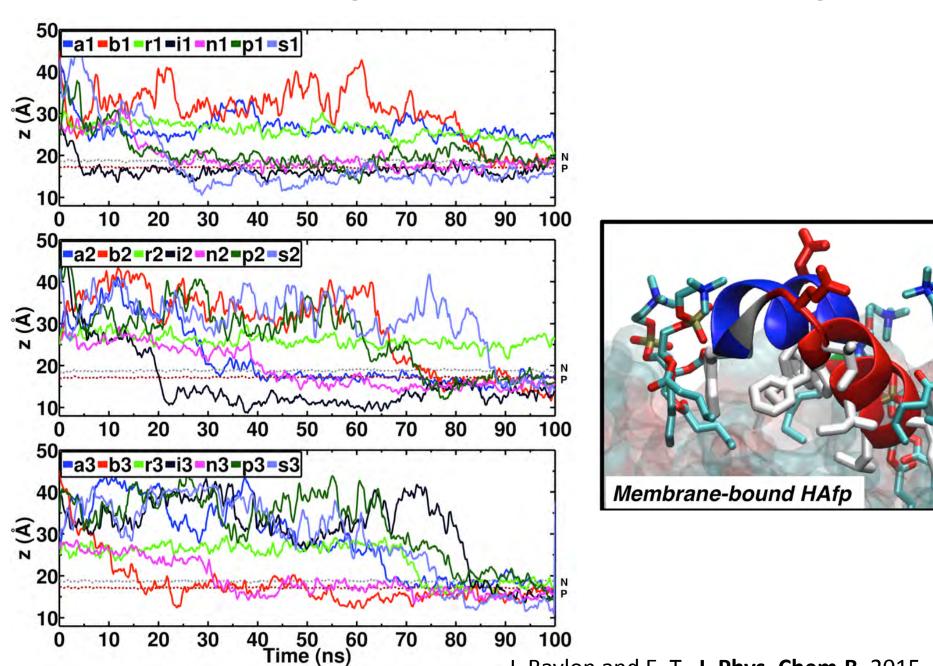
Membrane Binding of Influenza Hemagglutinin Fusion Peptide

Spontaneous binding observed in the majority of the simulations: 21 independent simulations starting from 7 different orientations



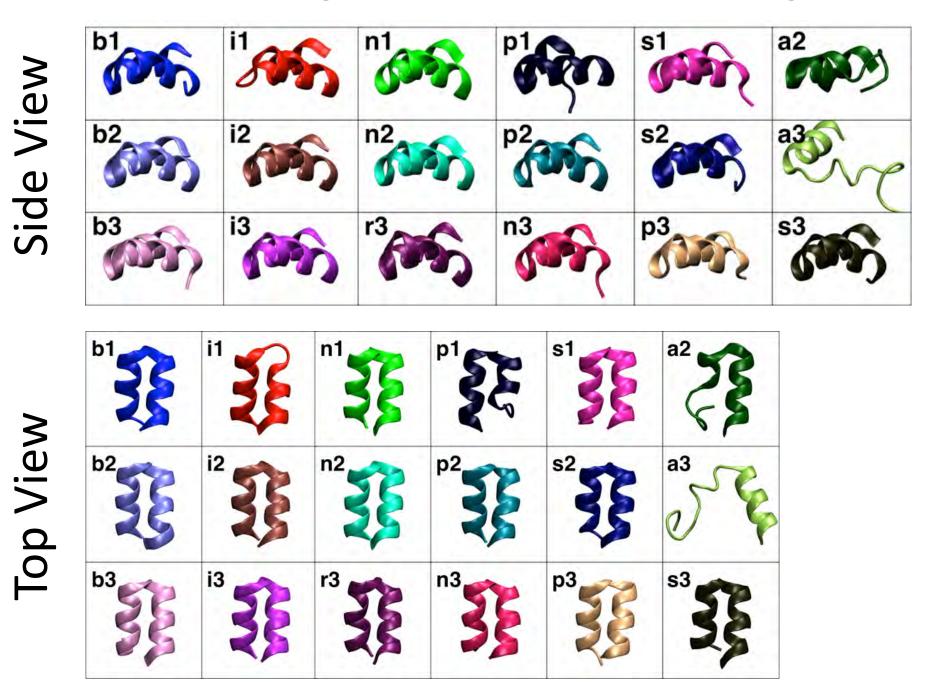
J. Baylon and E. T., J. Phys. Chem.B, 2015.

Remarkable convergence of membrane binding simulations



J. Baylon and E. T., J. Phys. Chem.B, 2015, in press.

Remarkable convergence of membrane binding simulations



J. Baylon and E. T., J. Phys. Chem.B, 2015.

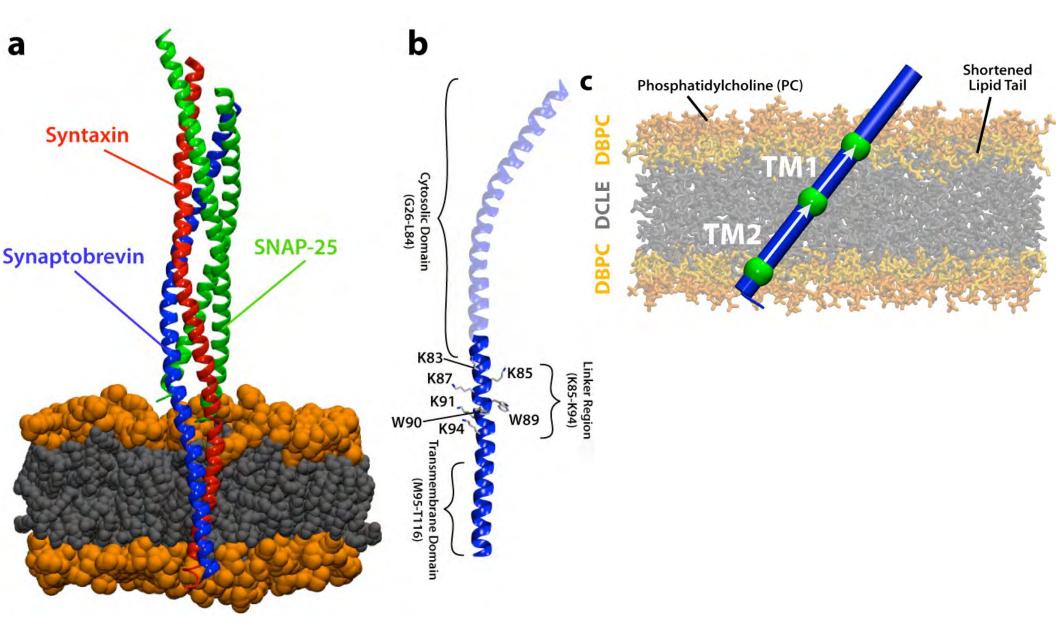
Robust Tilting of the Anchor Domain in Snare Protein Synaptobrevin



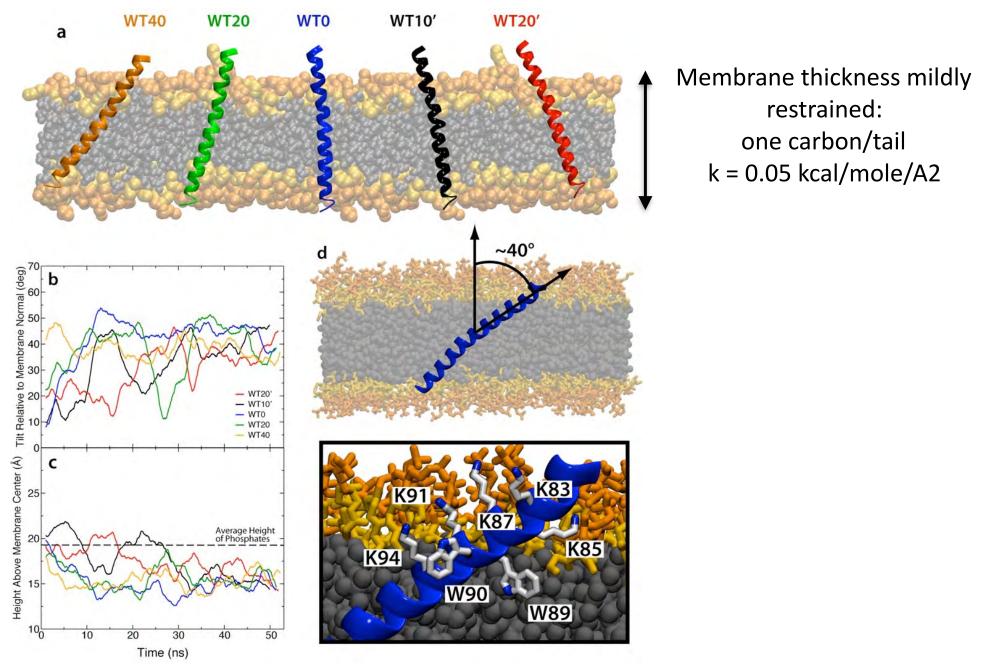


Mark Arcario

Andrew Blanchard

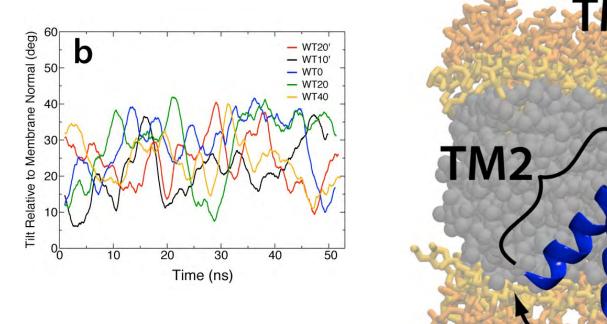


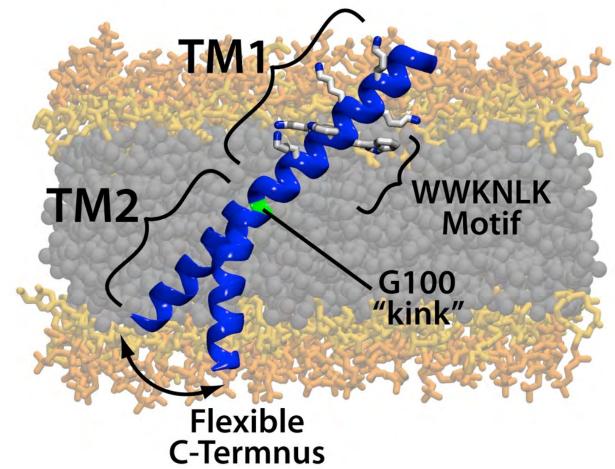
Robust Tilt Observed in Synaptobrevin



A. Blanchard*, M. Arcario*, K. Schulten, and ET, **Biophys. J.**, 107: 2112–21 (2014)

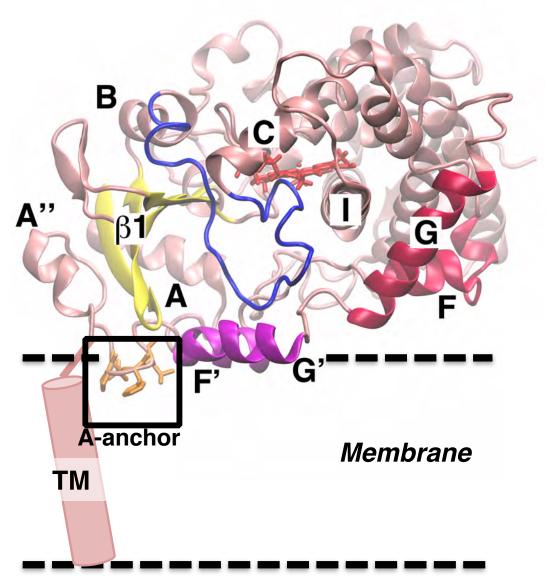
Identifying a Hinge





A. Blanchard*, M. Arcario*, K. Schulten, and ET, Biophys. J., 107: 2112–21 (2014)

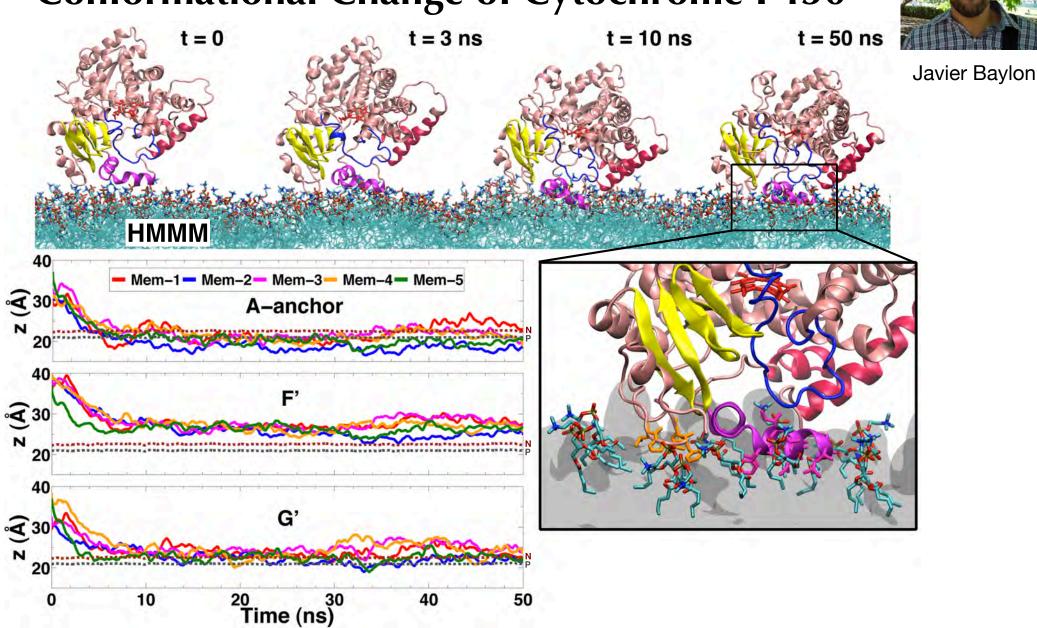
Cytochrome P450 3A4 (CYP3A4)



Yano et al., J Biol Chem, 279: 38091-38094, 2004

- Enzymes essential for the metabolism of xenobiotics and other compounds, found in all domains of life.
- In the human body, CYPs are membrane-bound proteins.
- The interaction with membrane mediates binding of substrates.
- in the human body, metabolizes about 50%- 60% of drugs that are metabolized in the body.

Insertion and Membrane-Induced Conformational Change of Cytochrome P450

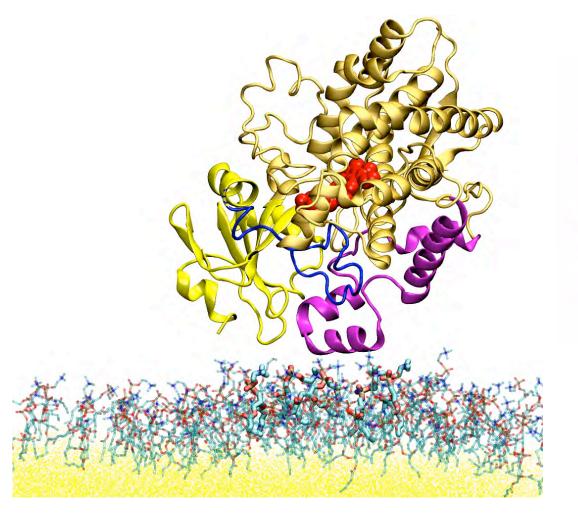


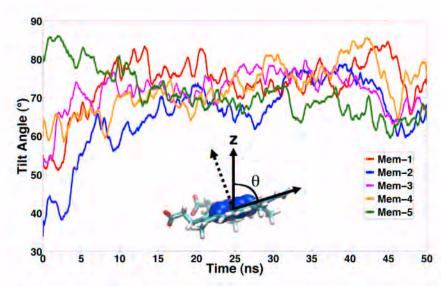
J. Baylon, I. Lenov, S. Sligar and ET, **JACS**, 135: 8542–8551 (2013)

Insertion and Membrane-Induced Conformational Change of Cytochrome P450



Javier Baylon





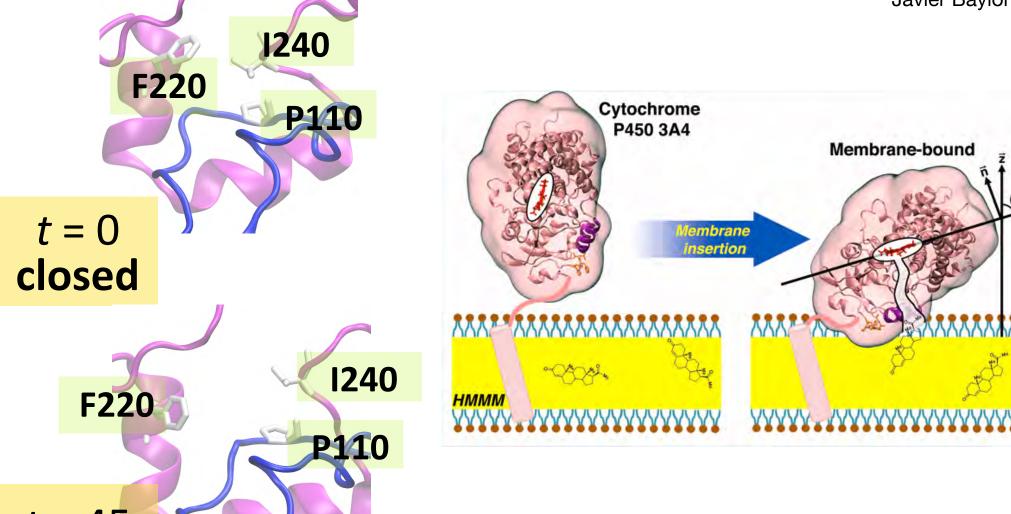
Within 10 degrees of experimental measurement of the tilt angle (S. Sligar)

J. Baylon, I. Lenov, S. Sligar and ET, JACS, 135: 8542–8551 (2013)

Insertion and Membrane-Induced Conformational Change of Cytochrome P450



Javier Baylon



t = 45 **open**

J. Baylon, I. Lenov, S. Sligar and ET, JACS, 135: 8542–8551 (2013)